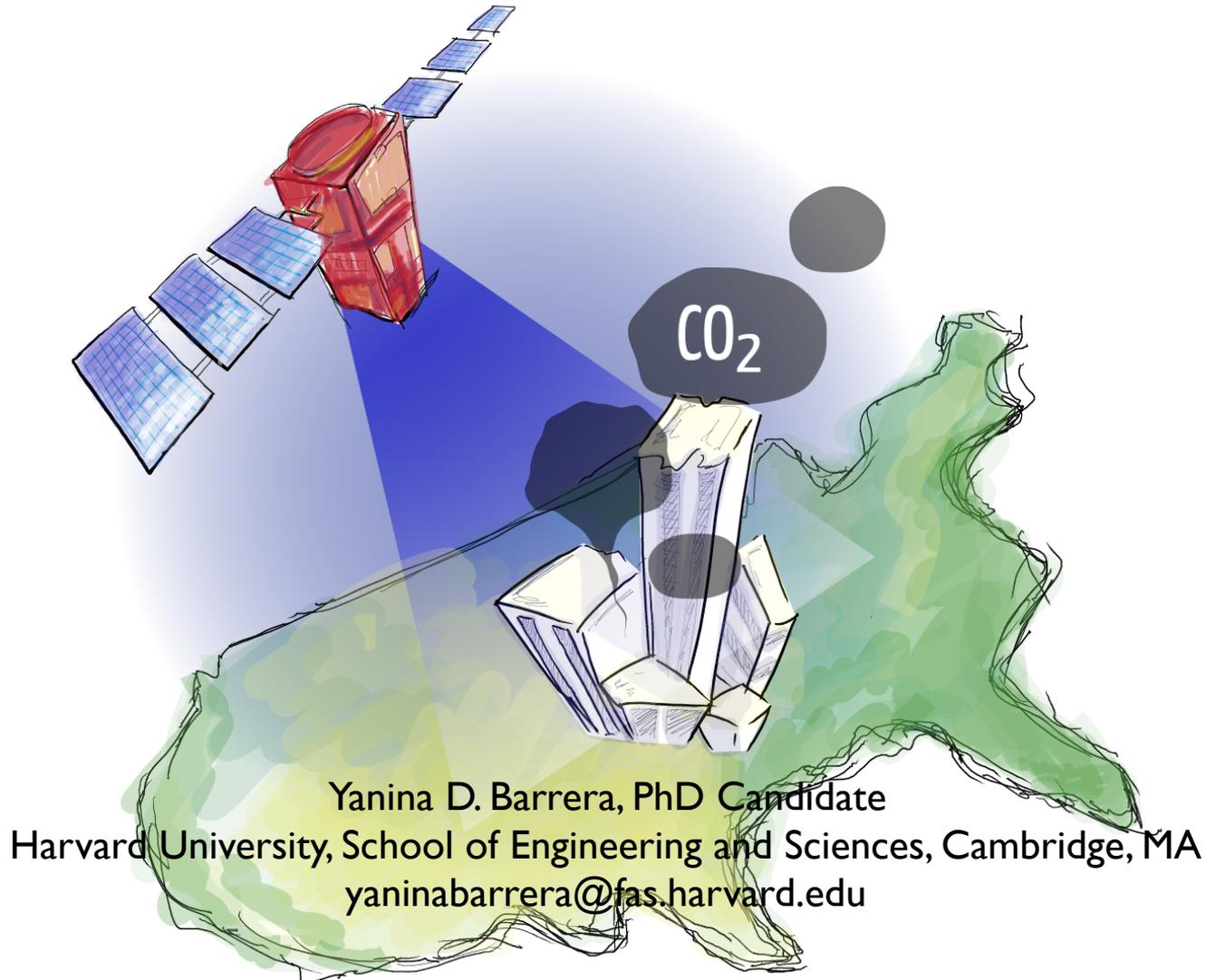


Air Pollution Knows No Boundaries



Yanina D. Barrera, PhD Candidate

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yaninabarrera@fas.harvard.edu

What air pollutants do you think are of major concern, when combusting coal?

- Carbon dioxide!
- Particulate matter
 - Toxic metals such as mercury compounds
- Sulfur dioxide emissions
- Nitrogen dioxide emissions

Pause for Thought

What events prompted air quality regulations?

How did we get here?

Timeline of Air Pollution Events & Regulation

1948 Donora Smog



control of hazardous
air pollutants

U.S.
Air Pollution
Control Act of
1955

U.S.
Clean Air Act
Amendments of
1967

Controlling
stationary &
mobile sources

U.S.
Clean Air Act
Amendments of
1990

1952 London's Great Smog



UK's
Clean Air Act of
1956

USA
Clean Air Act
Amendments of
1970

EPA established

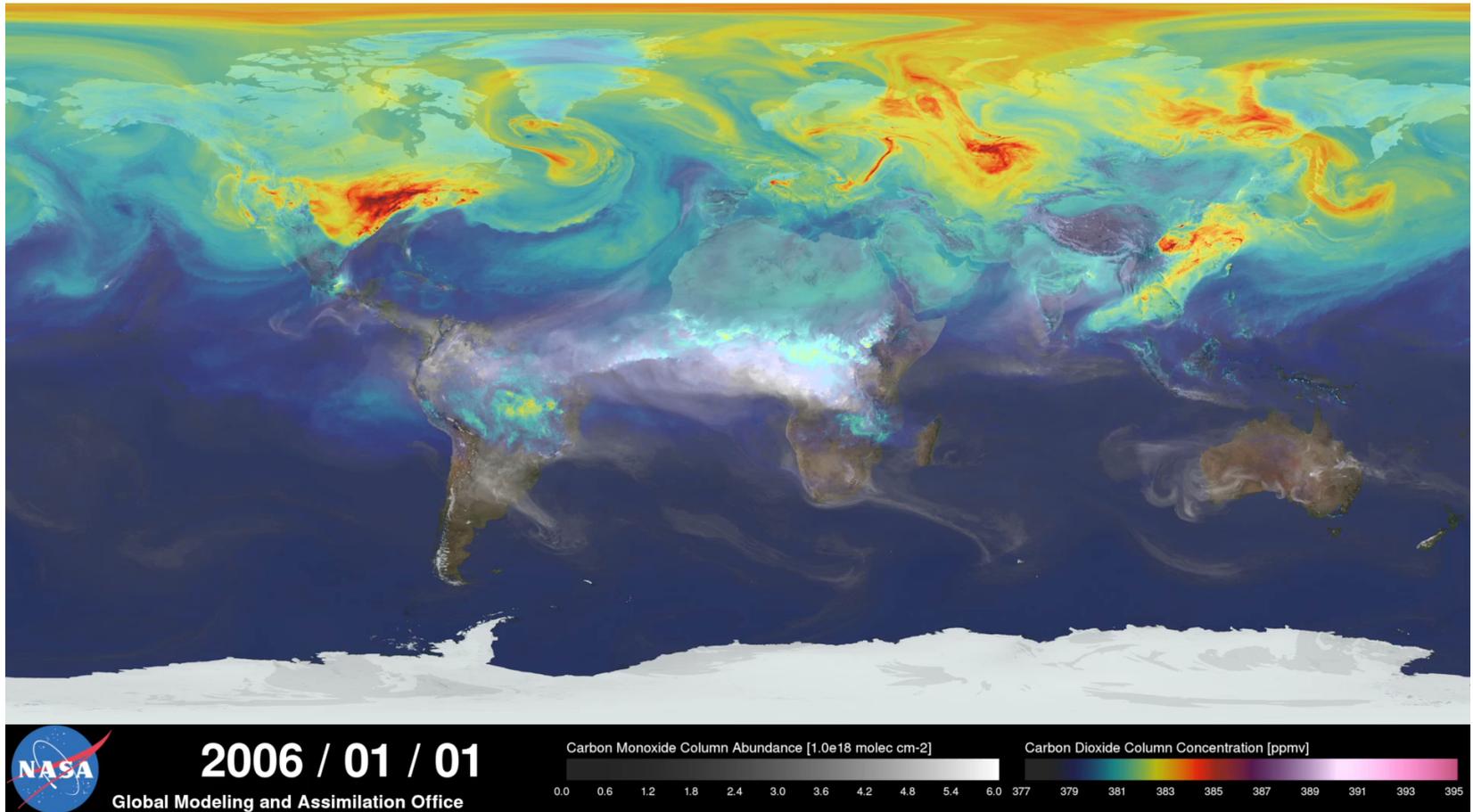
The Paris Climate Agreement

We will take action!



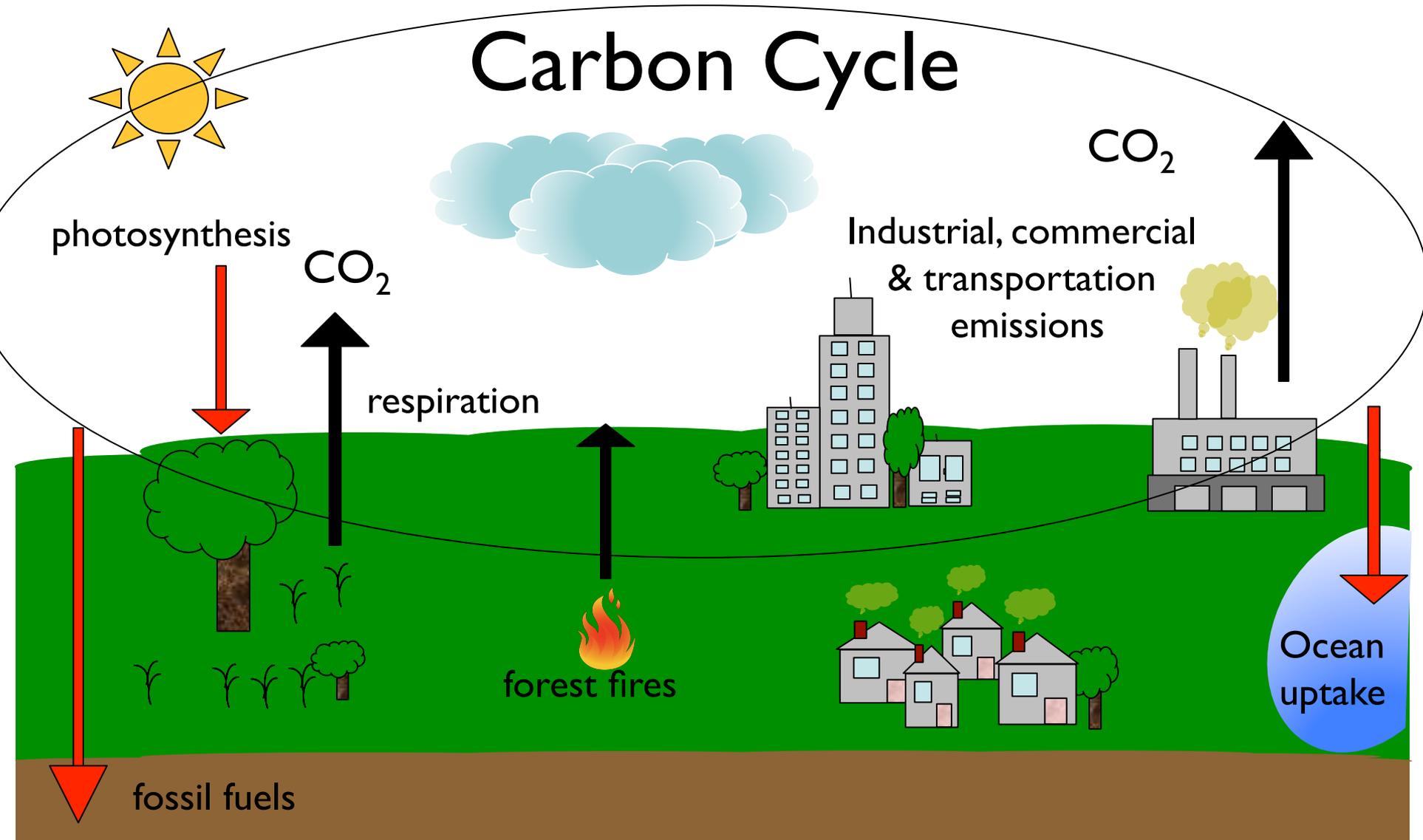
Source: New York Times, Christophe Petit Tesson/European Pressphoto Agency

But this problem is global!



Source: Bill Putman, NASA Goddard

Carbon Cycle



Roadmap for the evening

- I. How do we use models to understand air pollution in cities?

- II. What cutting-edge technology do we use on the ground, in the air, and in space?

- III. What do these instruments and models tell us?

Roadmap for the evening

- I. How do we use models to understand air pollution in cities?
- II. What cutting-edge technology do we use on the ground, in the air, and in space?
- III. What do these instruments and models tell us?

What is a model?

- A model is not the real world but a human construct to help us understand real world systems
- “Model” means both the computer program and the concept
- Models are accepted if it finds supporting evidence
- Models and experiments (or observations) can work together in order to get a better understanding of what we are trying to learn and in order to evaluate uncertainties

Pause for thought

Google maps from: carmel ca to: big sur

Get directions My places

car bus walking bicycle

A carmel ca

B big sur

Add Destination - Show options

GET DIRECTIONS

Suggested routes

CA-1 S/Cabrillo Hwy 28.1 mi, 42 mins

Driving directions to Big Sur, CA 2D III

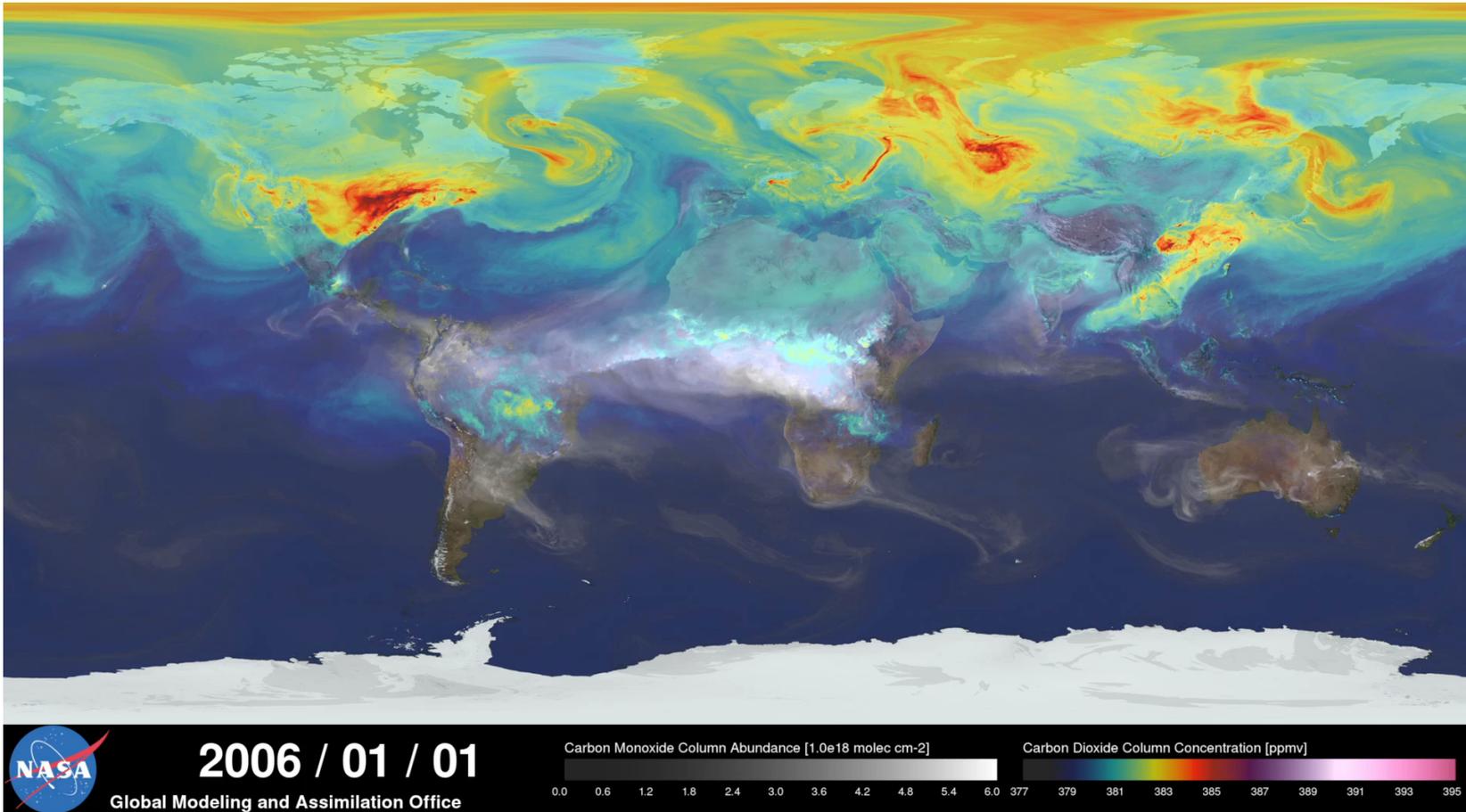
A Carmel, CA

1. Head east on Ocean Ave toward Dolores St 0.7 mi
2. Turn right onto CA-1 S/Cabrillo Hwy 27.1 mi
3. Slight left onto Pfeiffer Big Sur 0.2 mi
4. Take the 2nd left to stav on Pfeiffer Bia

Data LDEO-Columbia, NSF, NOAA
© 2011 Google
© 2011 Europa Technologies

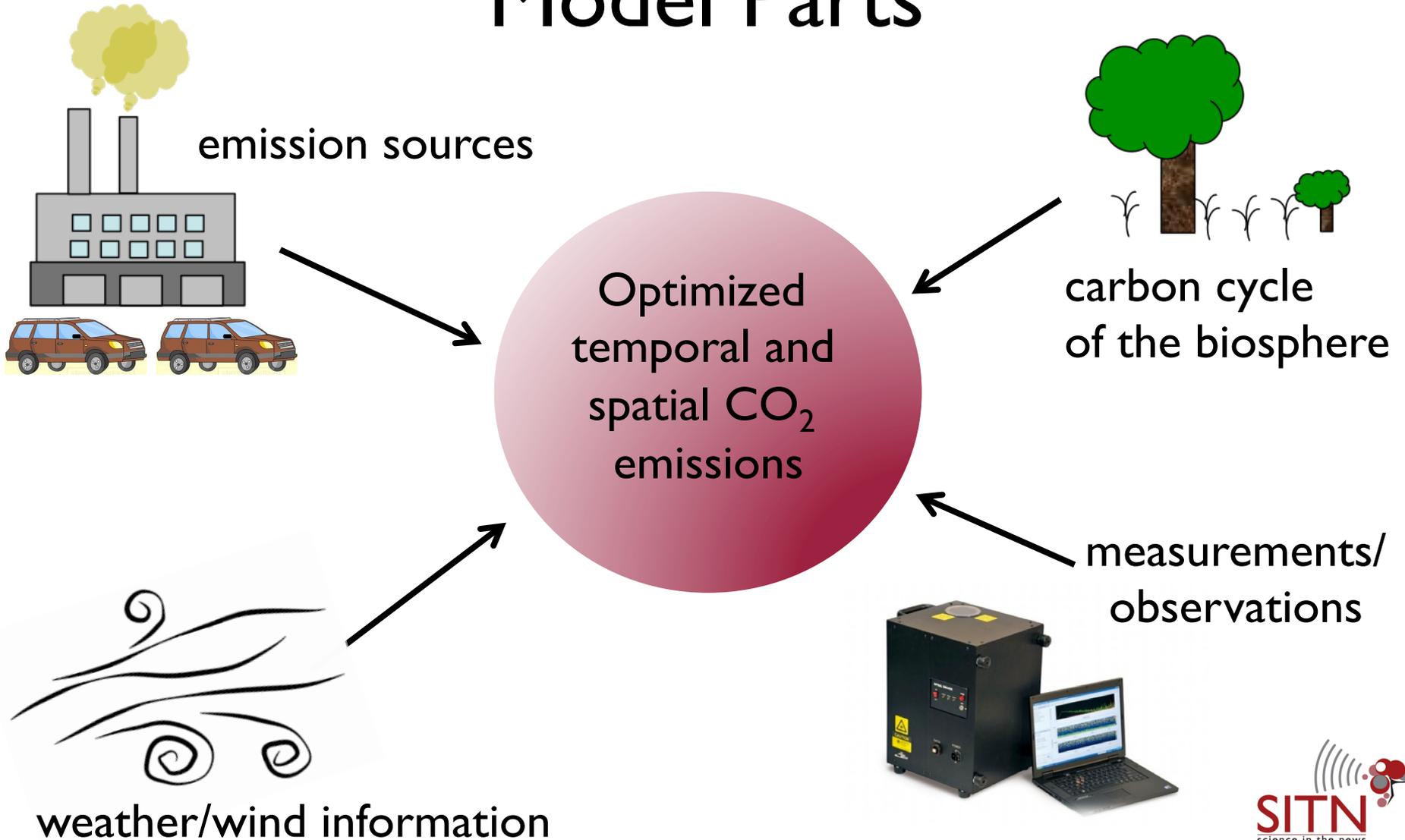
Google earth
Terms of Use

But how do we build models to study air pollution in cities?



Source: NASA Goddard Space Flight Center (bottom), NC university (top)

Model Parts

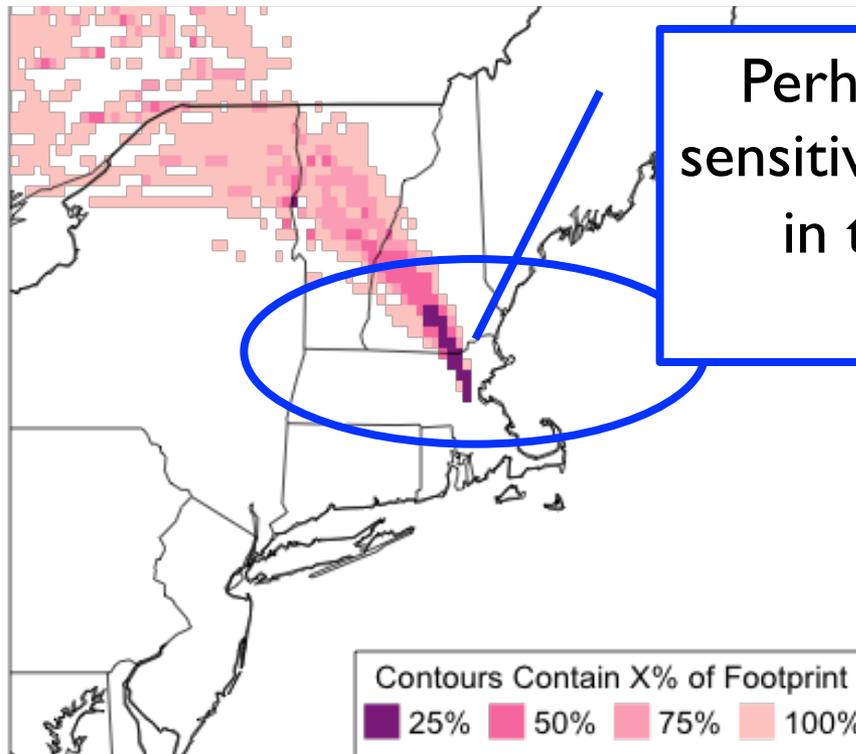


Model product showing sensitivity footprint

Footprint showing sensitivity of the *particles* that see surface observations

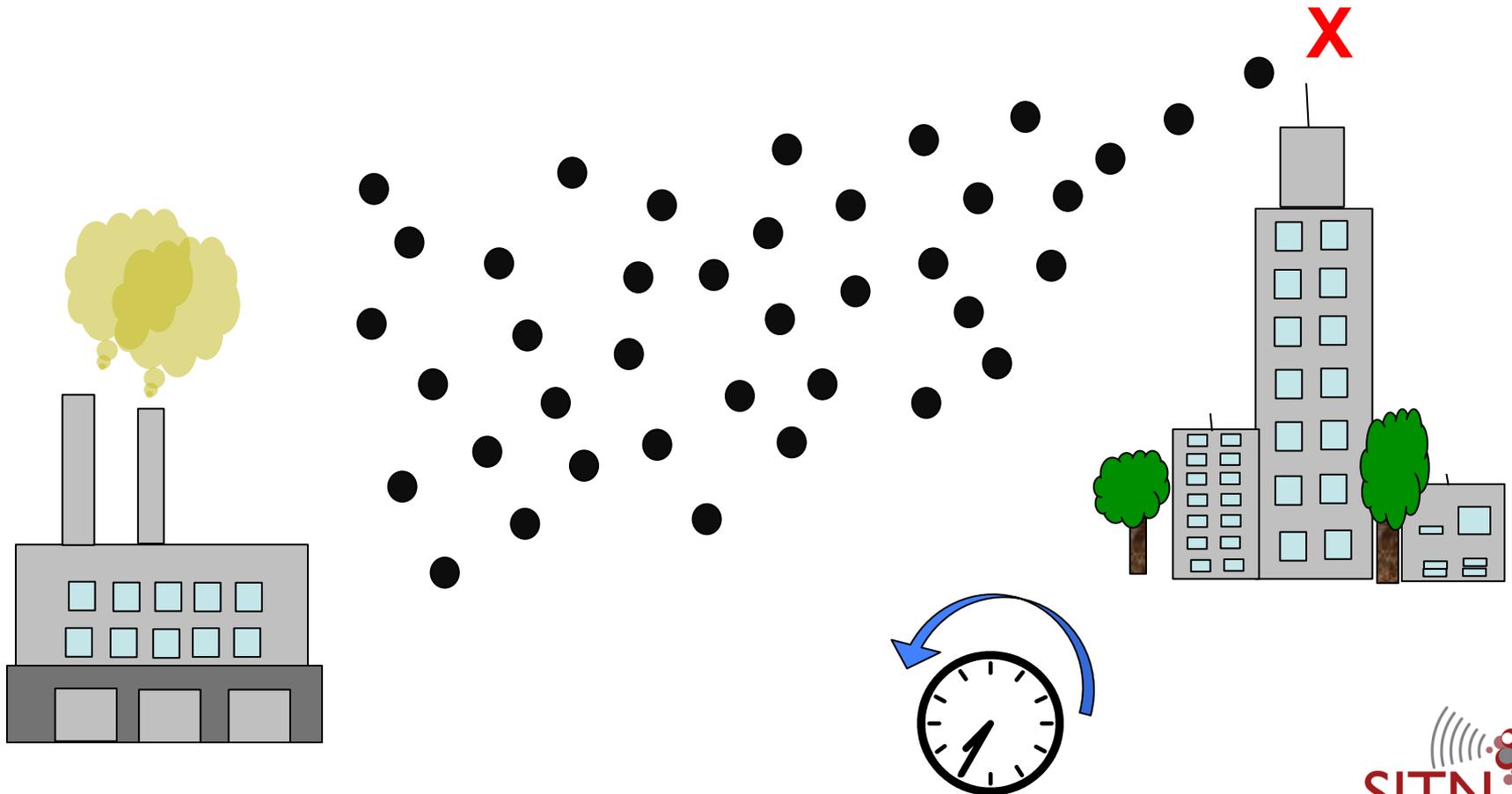
Date:
Nov. 11th, 2012

16 hours
backwards in
time



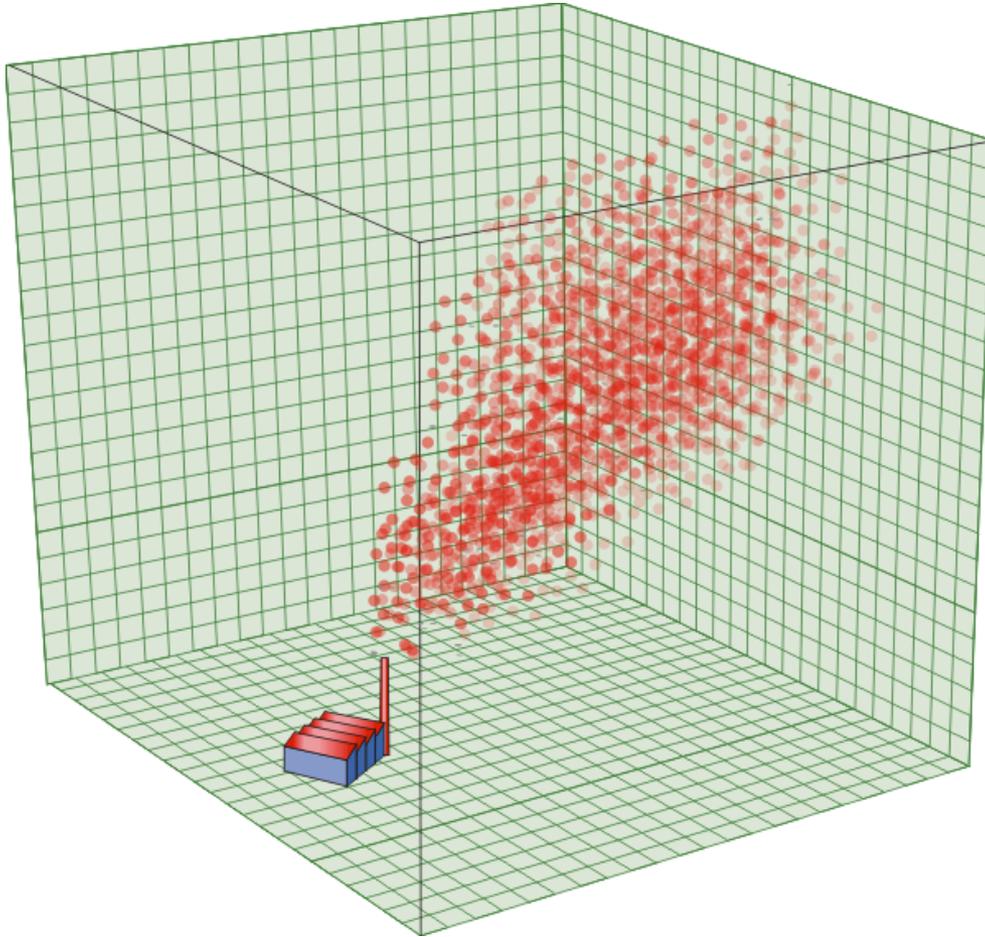
Perhaps particles are sensitive to wood burning in the wintertime?

We get these products by using a transport model!



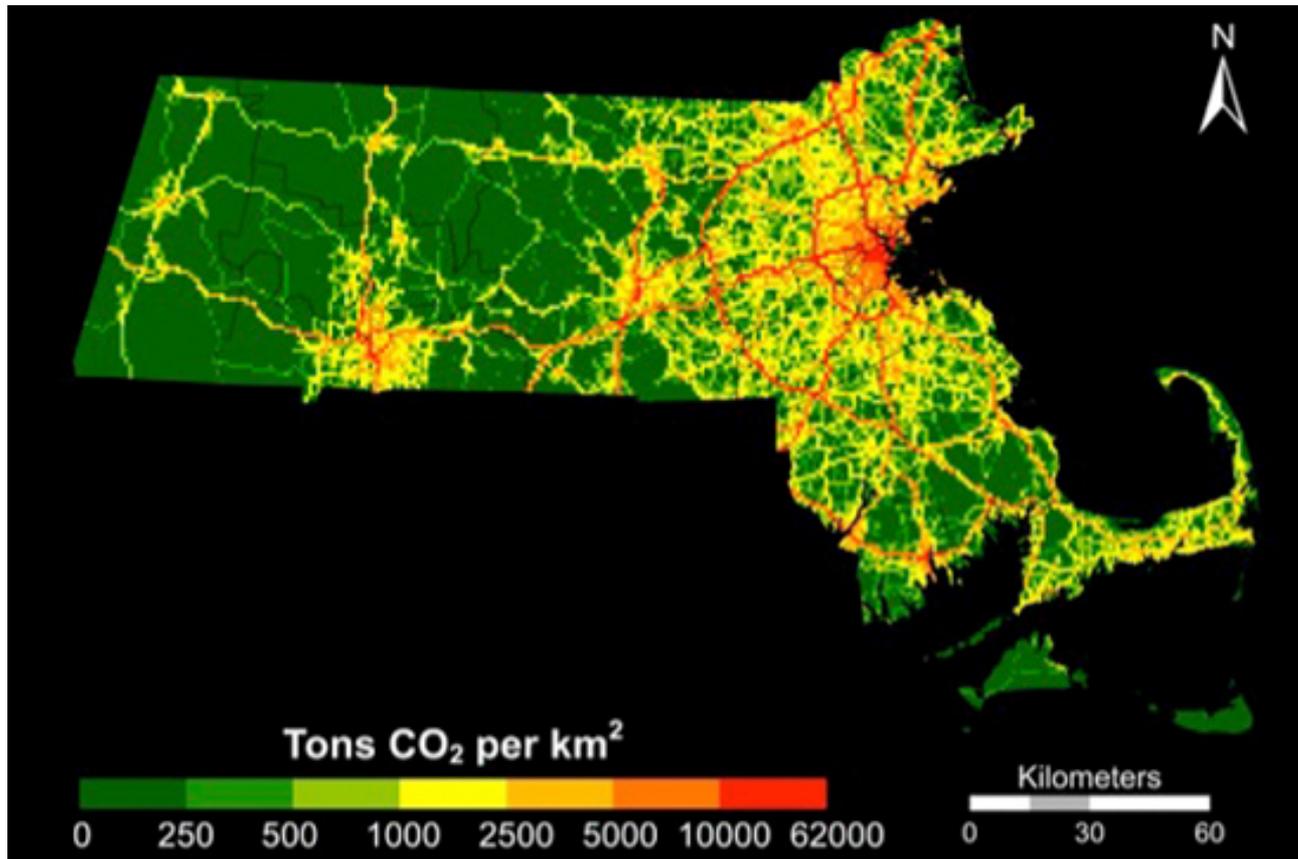
Backwards in time

3-D transport of particles



- We can track particles on a horizontal grid (map)
- We can track particles vertically in the atmosphere and identify air pollution sources most influencing CO₂ levels!

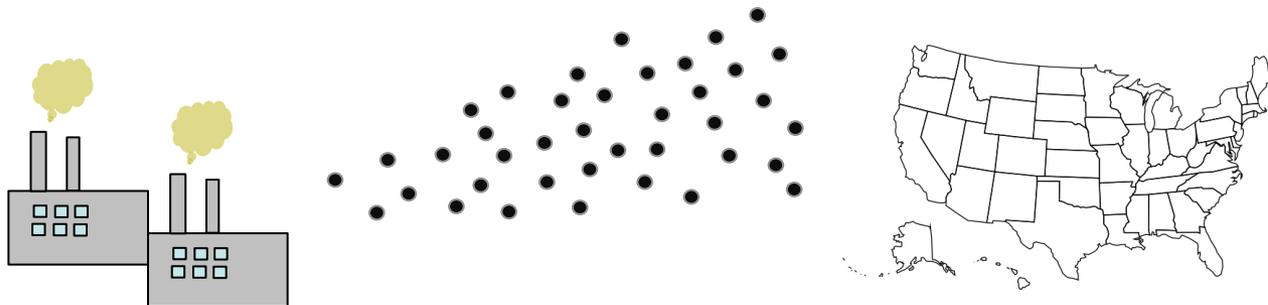
Integrate information about emission sources
→ on-road CO₂ emissions inventory



On-road transportation is responsible for 28% of all U.S. fossil-fuel CO₂ emissions

Part I Summary

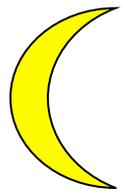
- Transport models are heavy computationally and expensive to run but helps us study air pollution locally and regionally, at the observation site in consideration
- Information of the winds can help us understand where the air pollution came from
- Emissions inventory helps us identify emissions sources as we track particles backwards in time on a map!



BREAK for Questions

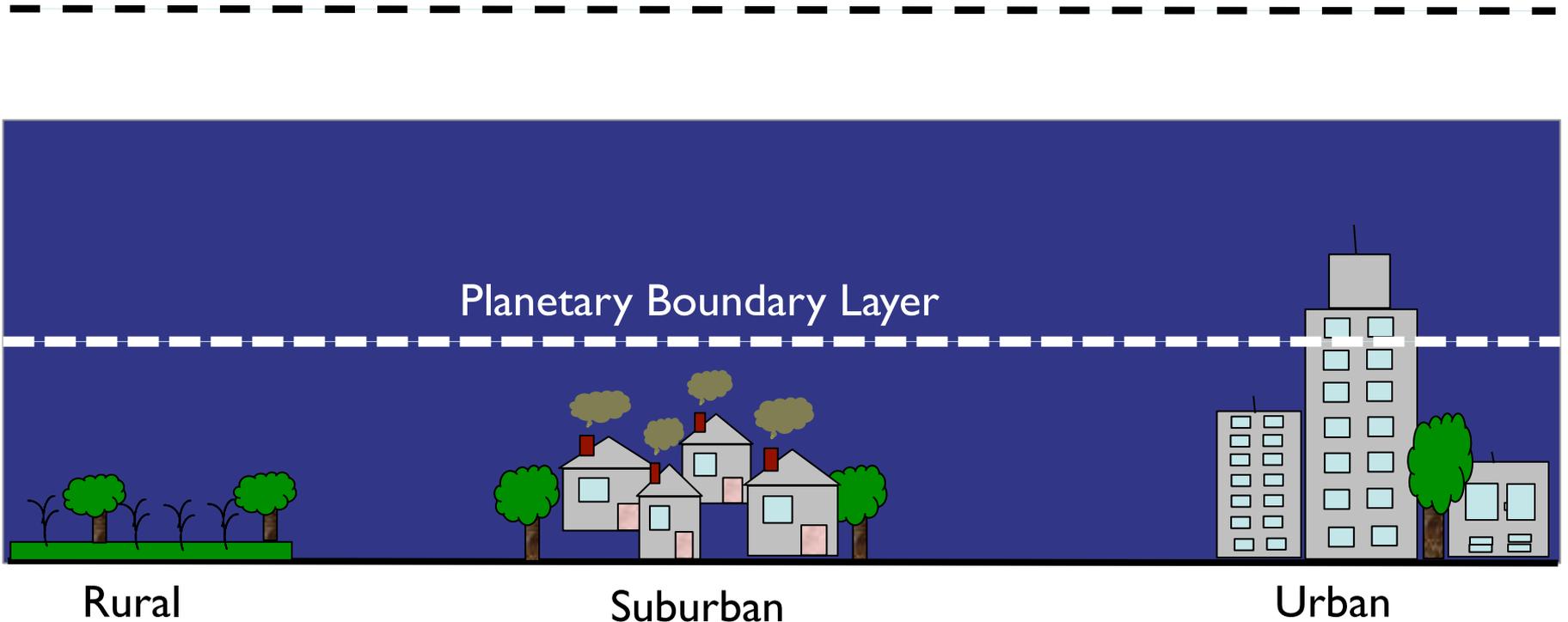
Roadmap for the evening

- I. How do we use models to understand air pollution in cities?
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The air we breath - nighttime

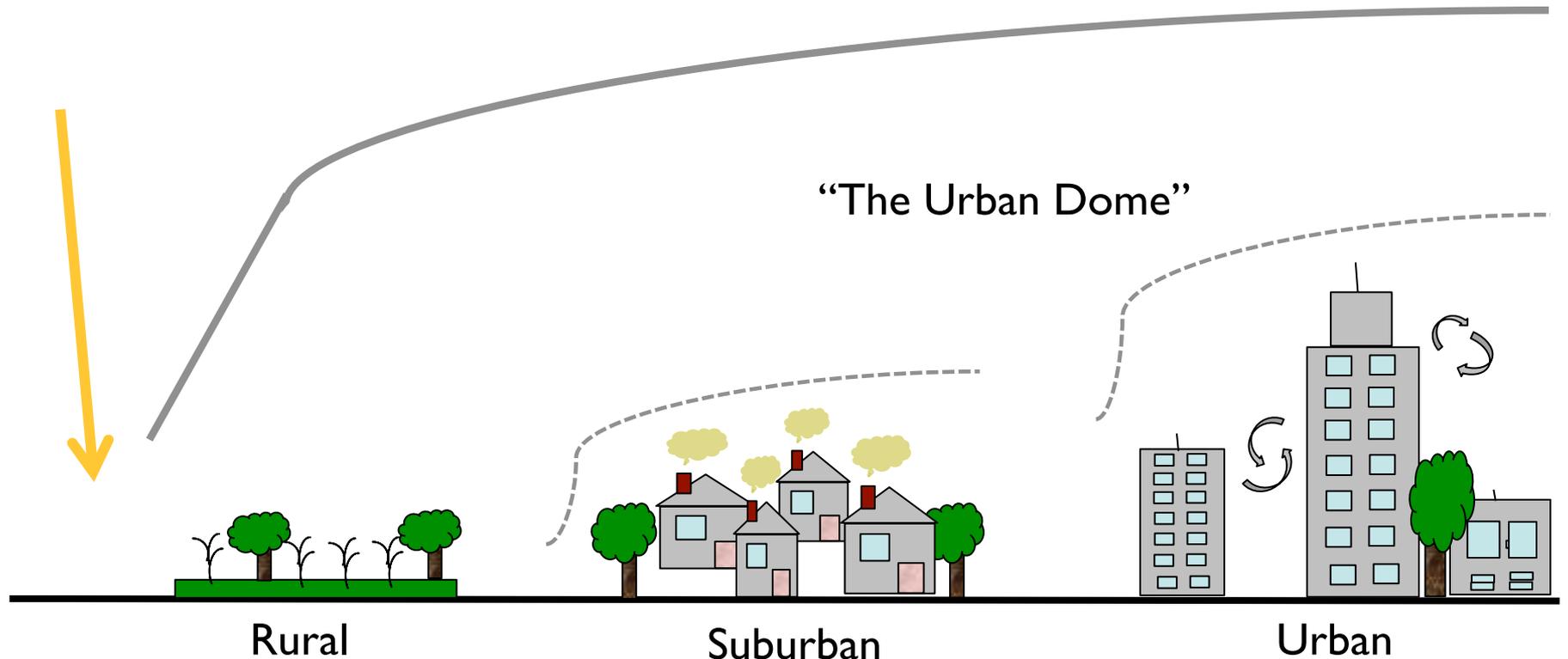
Planetary Boundary Layer (PBL)





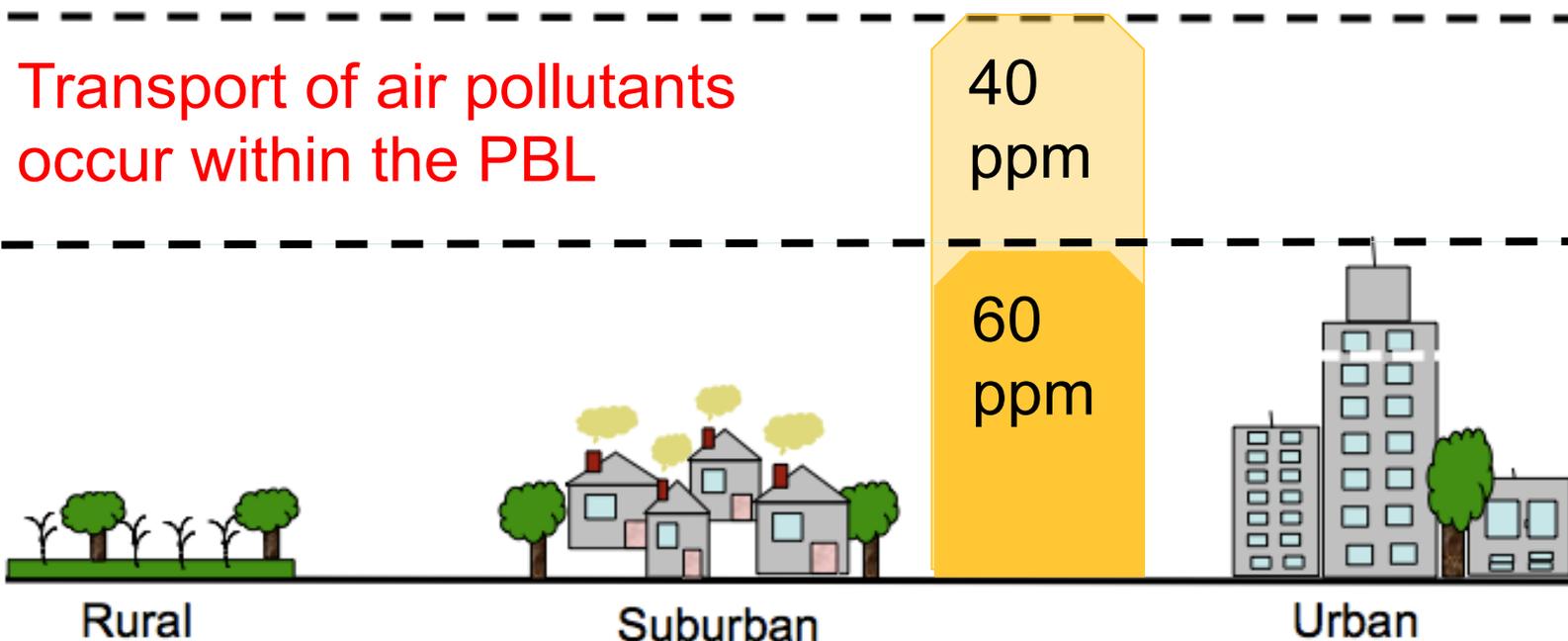
The air we breath - daytime

Planetary Boundary Layer (PBL)



Why do we care about the PBL?

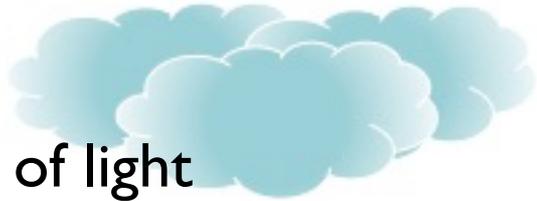
Height of PBL helps us determine our concentrations at the surface



How do we measure this layer?

Lidar (“Light Detection and Ranging”)
Technology

Lidar Technology Detects Backscattering!



return of light



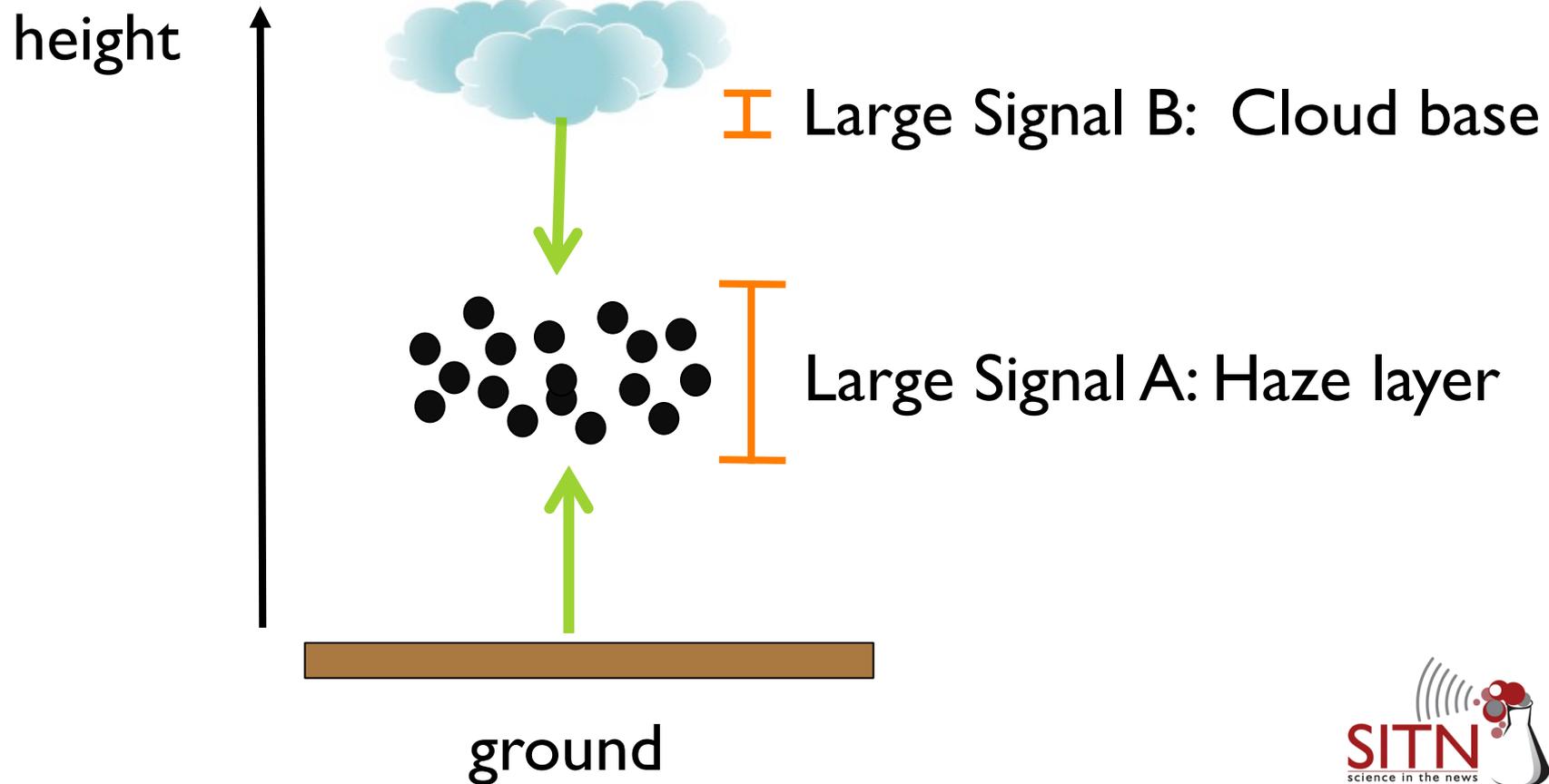
- Backscattering deflection of light particles at an angle of 180°

- Lidar can measure the “intensity” of backscatter at each height point in the atmosphere

pulse of light



one Lidar pulse = many returns



Lidar measures backscatter from the atmosphere



return of light



Question:

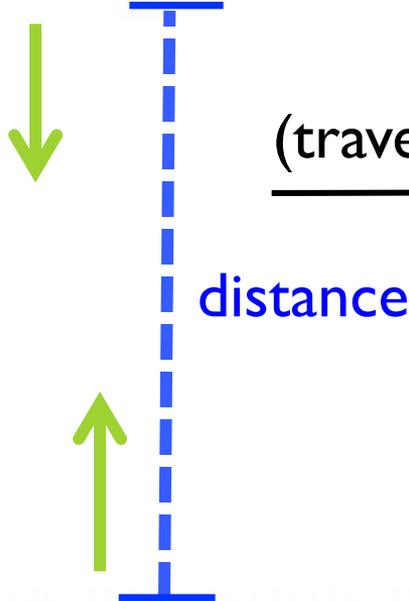
How do we determine the height of the bottom of the cloud with Lidar?



pulse of light



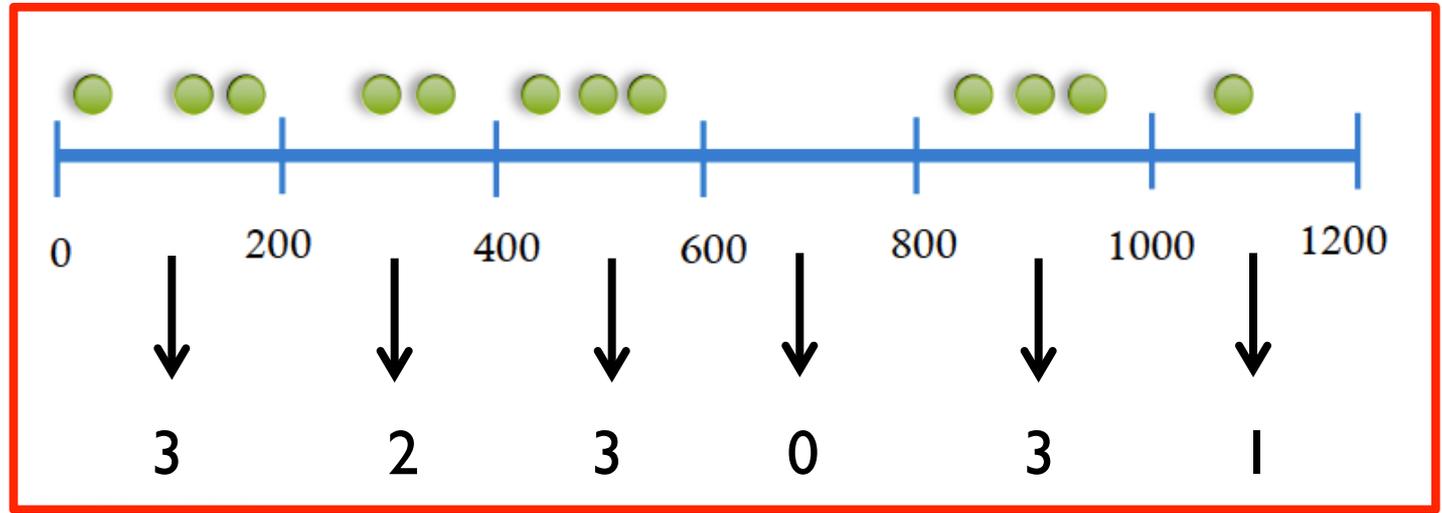
Calculate height of clouds or structures



$$\frac{(\text{travel time}) \times (\text{speed of light})}{2} = \text{height}$$

distance

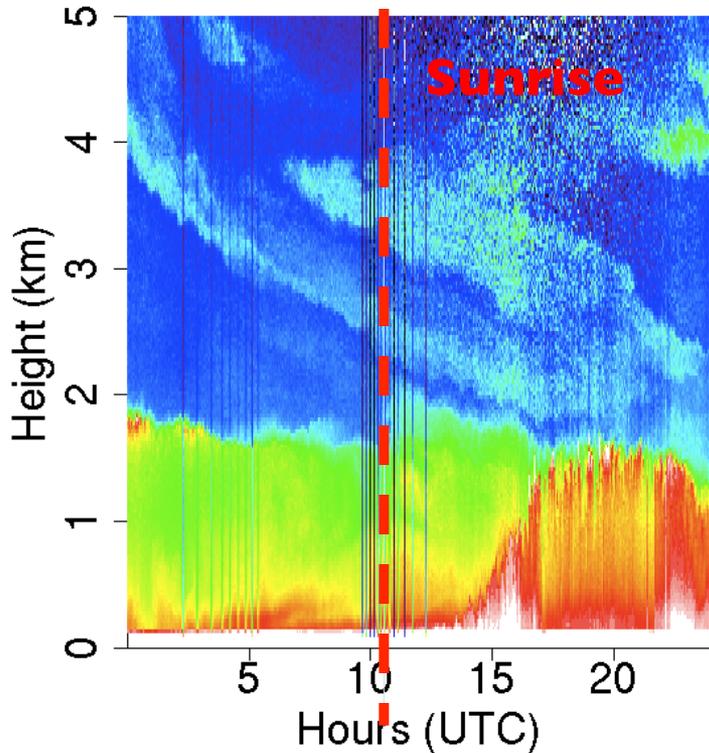
Lidar stores the amount of light returned as “photon” counts in bins every 200 nanoseconds



Computer program stores information and we download data to analyze

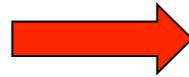
We can image atmospheric structures!

Aug. 21, 2013

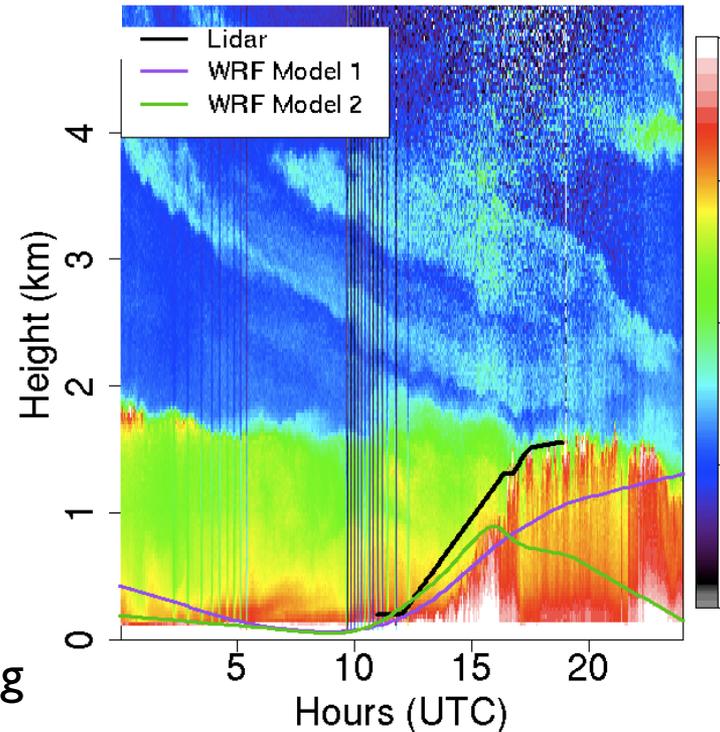


More backscattering
(more particles!)

Less backscattering
(fewer particles!)



Aug. 21, 2013



Great! We can use Lidar technology to measure the height of the Planetary Boundary Layer.....

But where do we measure carbon dioxide emissions?

Northeast Measurement Network

Copley
215 m (4 corners)
July 2012 - present



Harvard Forest
30 m (8 levels)
Aug 2012 - Present



Canaan, NH



Thompson Island, MA



Stockholm, NJ



Mineola, NY

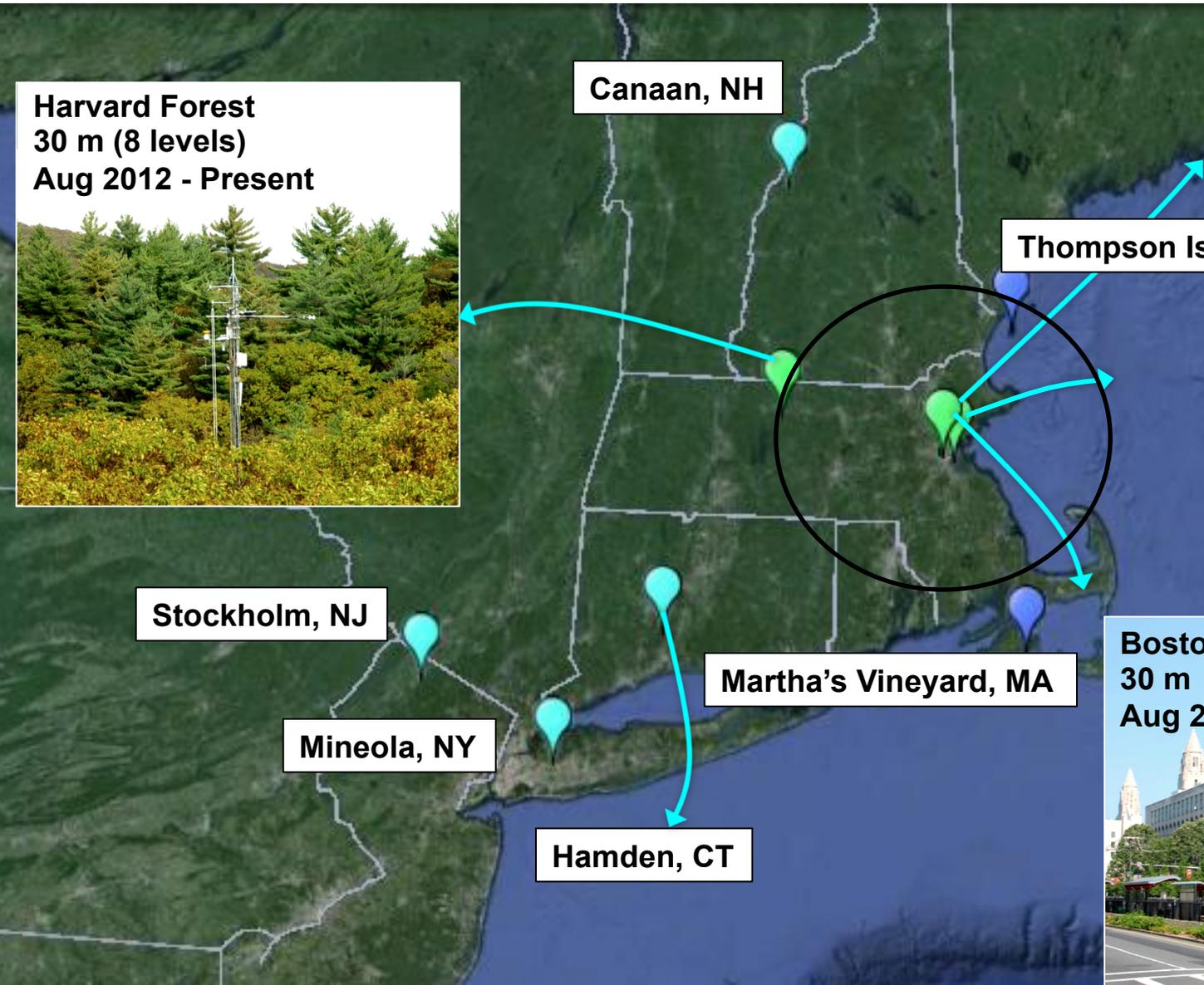


Hamden, CT

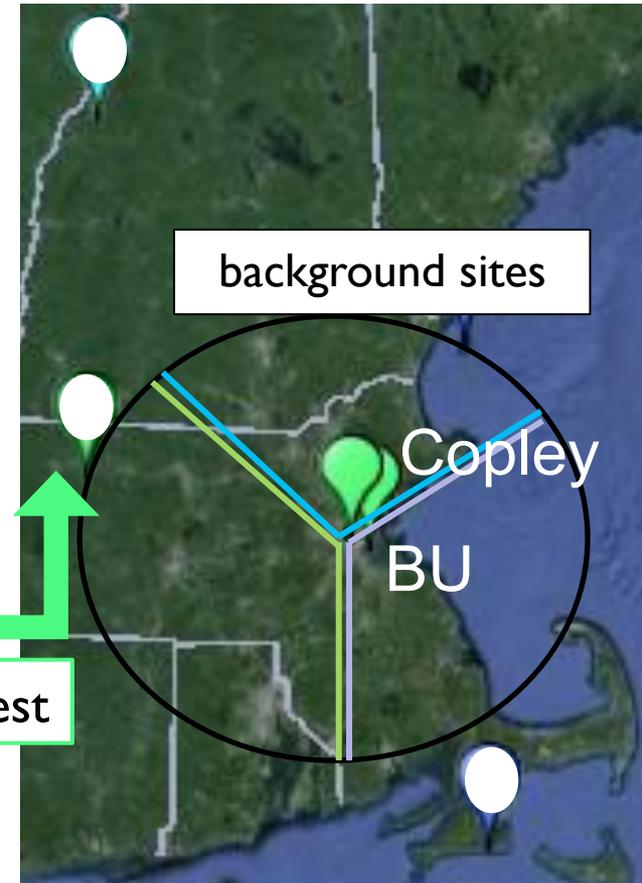
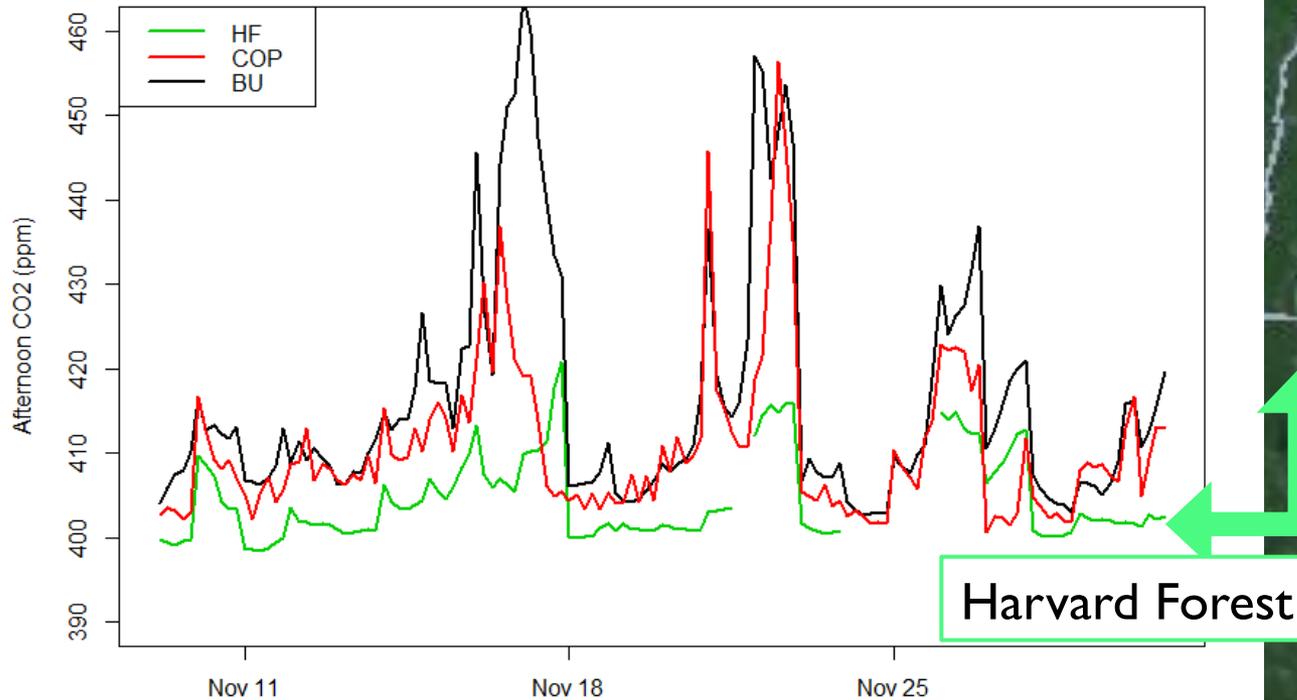


Martha's Vineyard, MA

Boston University
30 m
Aug 2012 - present



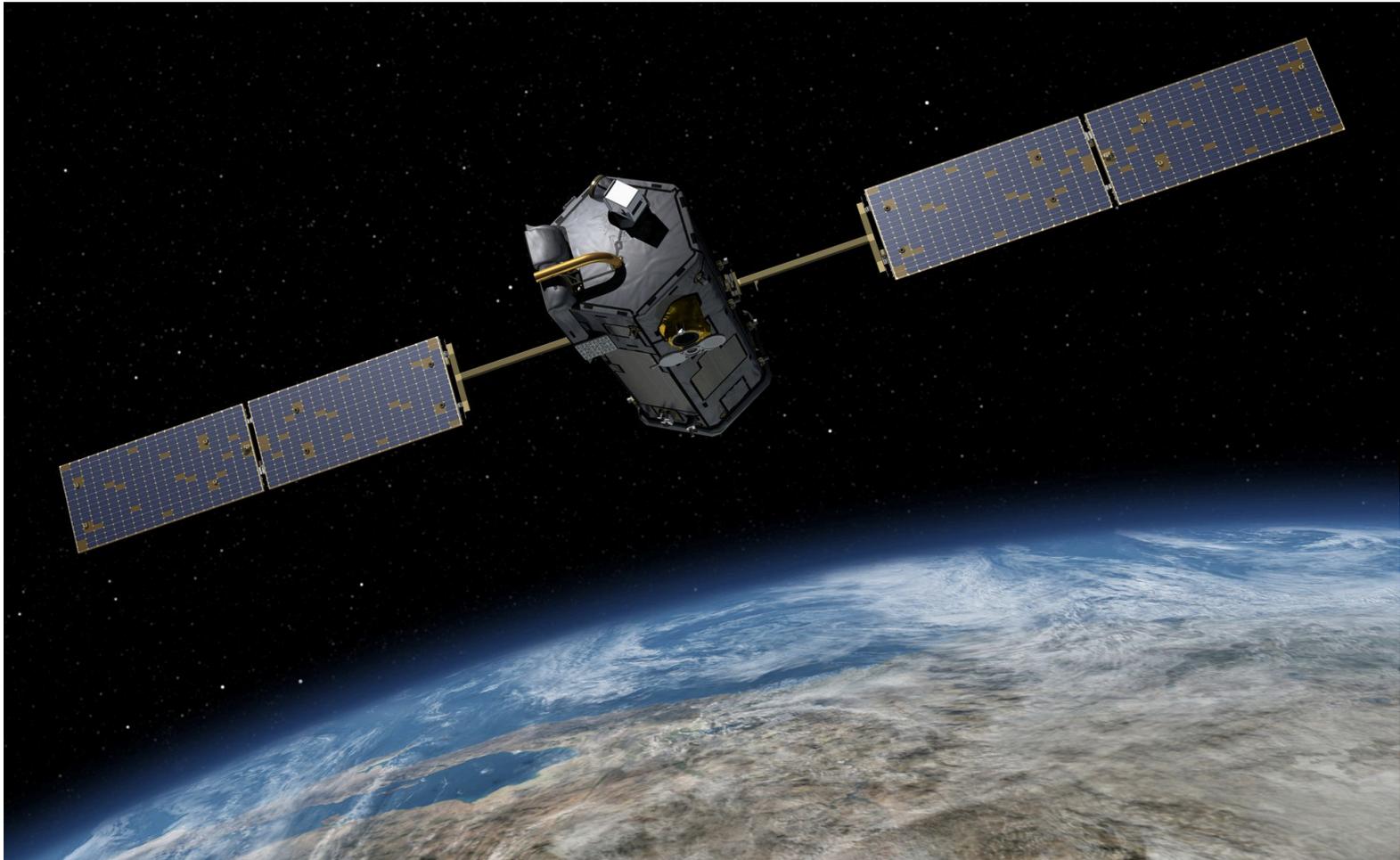
CO₂ results from Earth Networks



Good “background” to compare “enhancements” in CO₂ observations



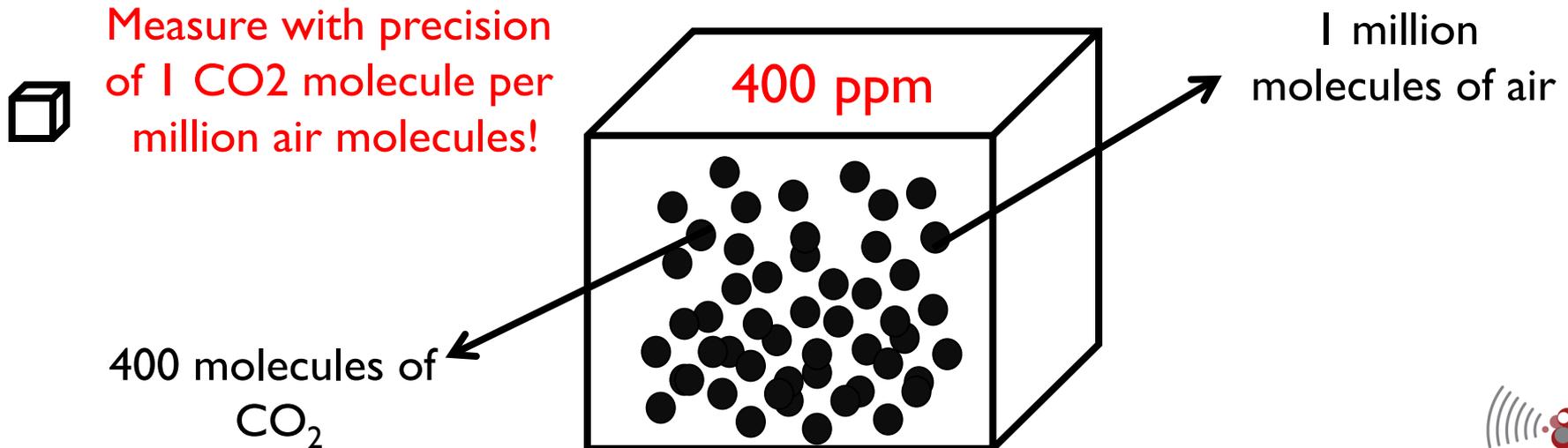
Orbiting Carbon Observatory 2 (OCO-2) NASA Satellite



Source: NASA's Jet Propulsion Laboratory

How well can OCO-2 measure carbon dioxide levels on earth?

- OCO-2 measures carbon dioxide levels with a precision of about 1 part per million, or ppm. What does that mean?
- Today's carbon concentration, is 400 ppm
 - The highest it has been in at least 800,000 years!



What does one ppm mean to us?

- Recent monthly average at Mauna Loa, Hawaii, observatory

~ 4 ppm difference in 1 year!

February 2016: 404.02 ppm
February 2015: 400.26 ppm

Last updated: March 7, 2016

BREAK for Questions

BREAK for Intermission

Roadmap for the evening

- I. Cutting edge technology on the ground, in the air, and in space!
- II. How do we use models to understand air pollution in cities?
- III. What do these instruments and models tell us?

Northeast Measurement Network

Copley
215 m (4 corners)
July 2012 - present



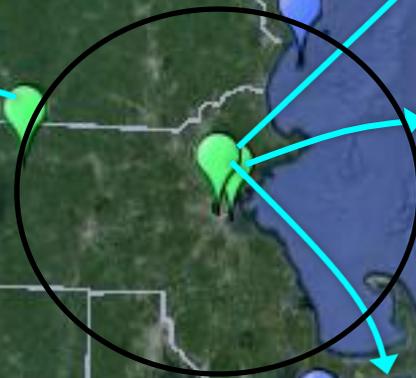
Harvard Forest
30 m (8 levels)
Aug 2012 - Present



Canaan, NH



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Mineola, NY



Martha's Vineyard, MA

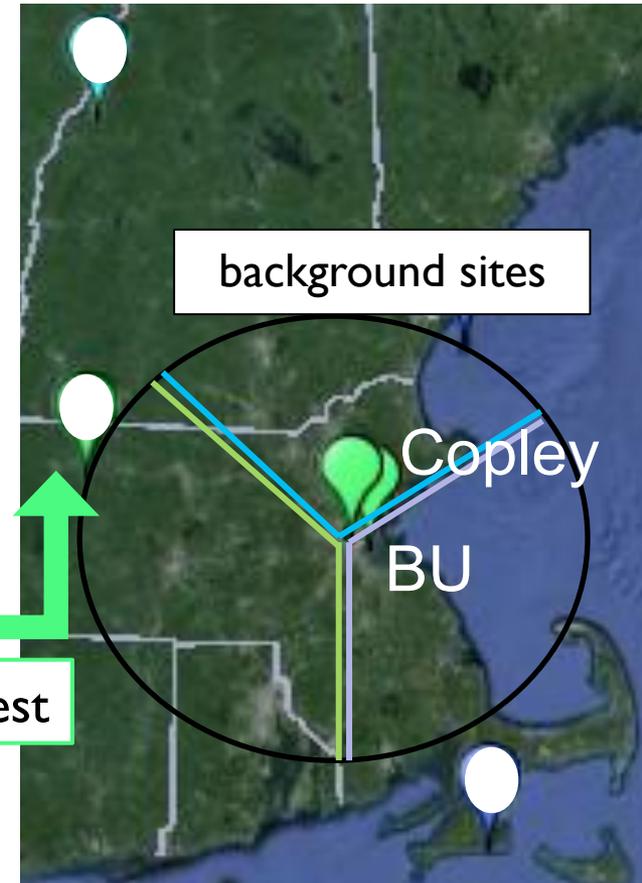
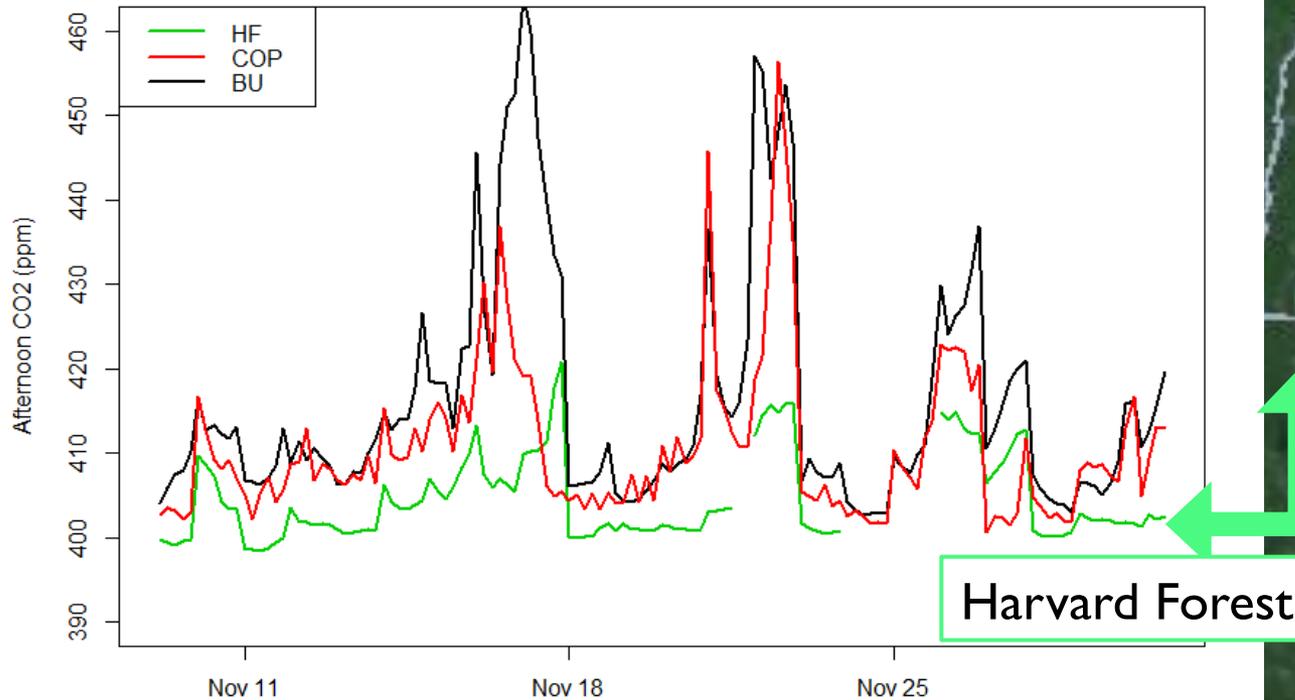


Hamden, CT

Boston University
30 m
Aug 2012 - present



CO₂ results from Earth Networks



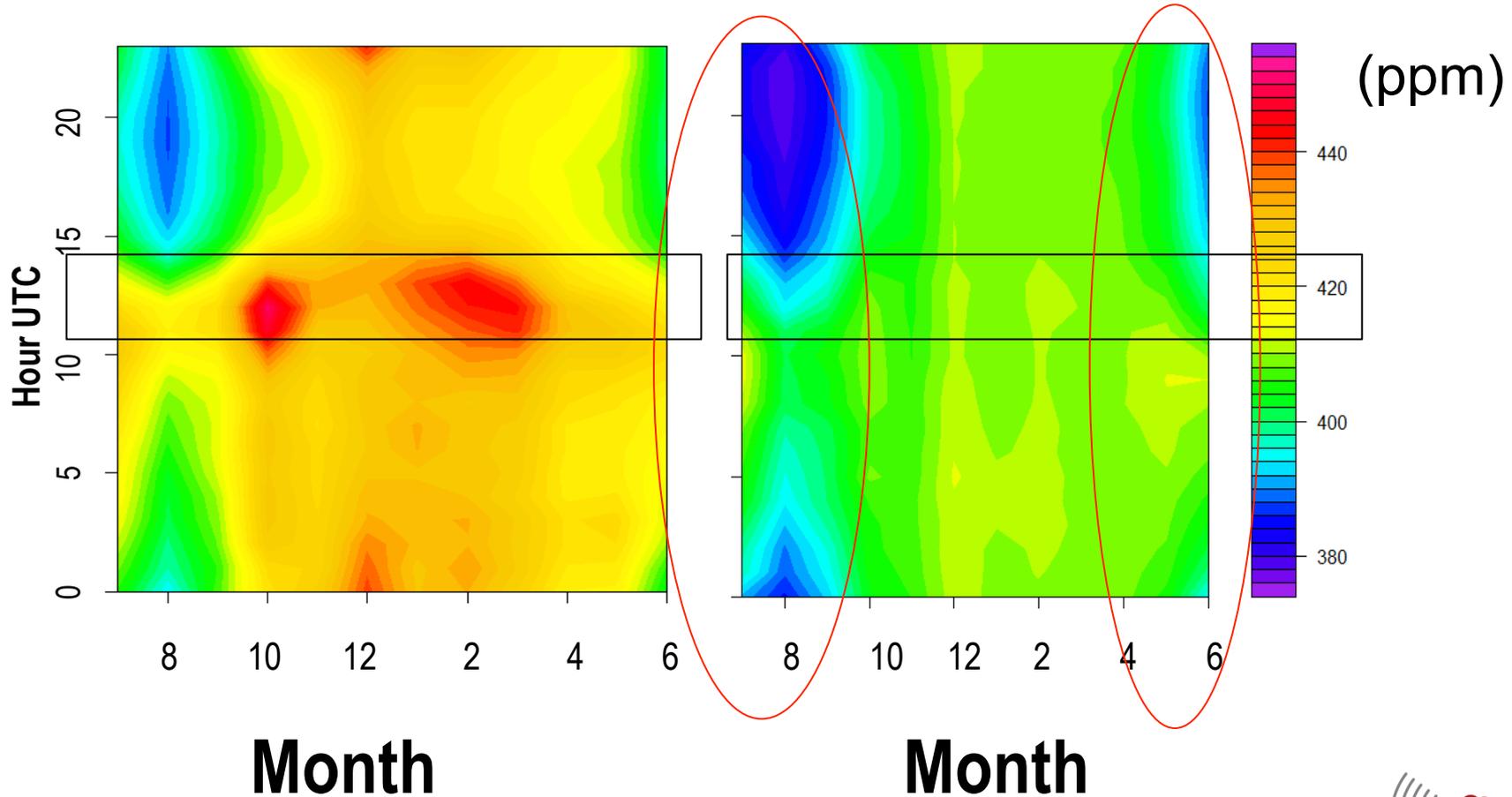
Good “background” to compare “enhancements” in CO₂ observations



CO₂ concentration results

Boston University

Harvard Forest

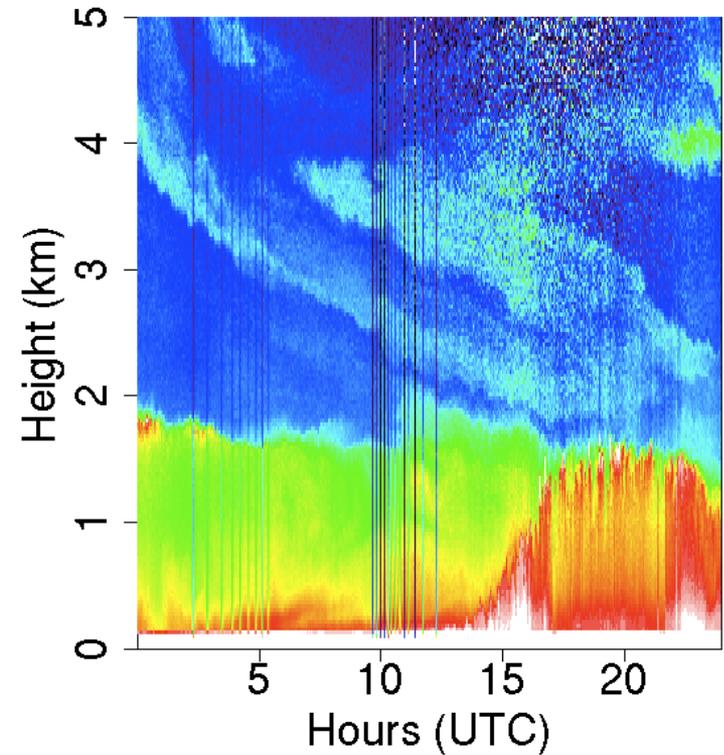


We can image atmospheric structures!

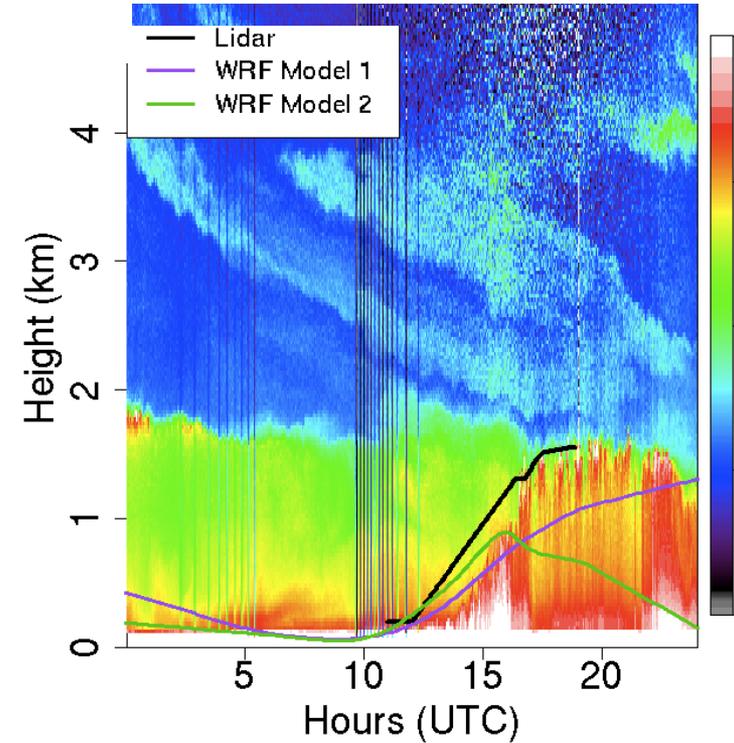
20130821 NRB Profile

More backscattering
(more particles!)

20130821 NRB Profile

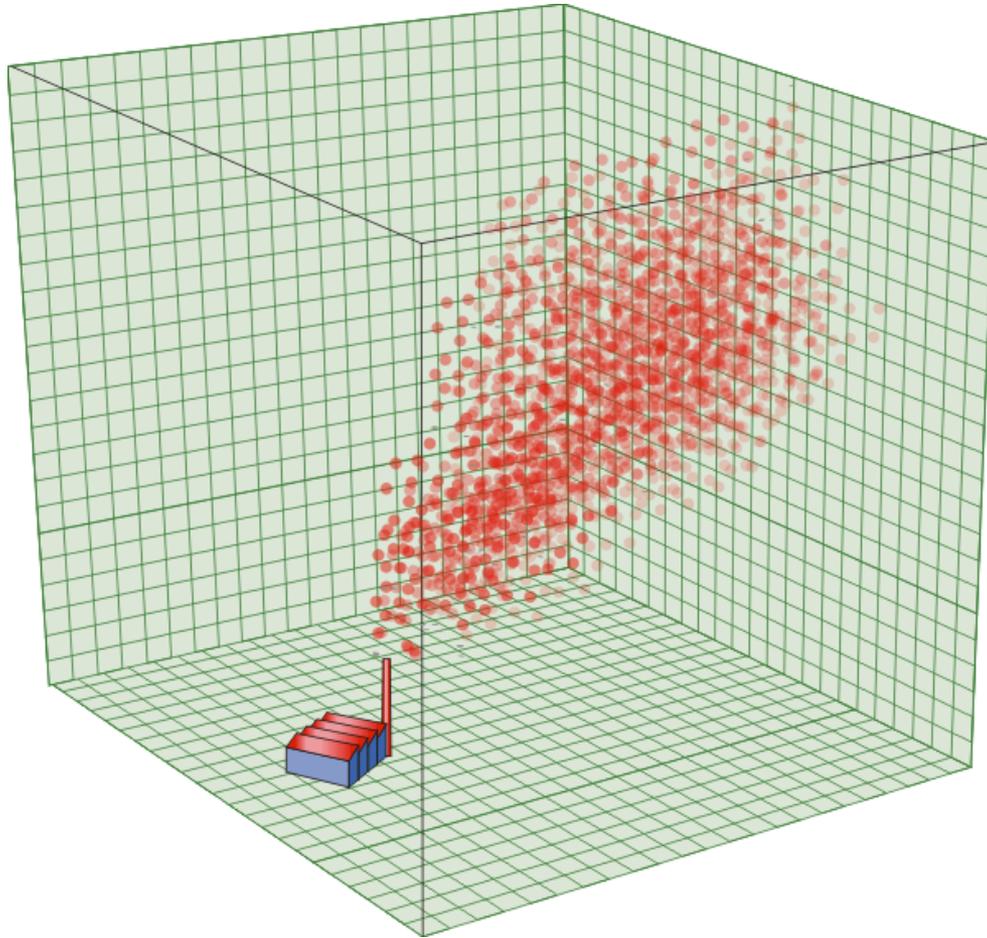


Less backscattering
(less particles!)



What if the PBL height does not agree with
our Lidar observations?

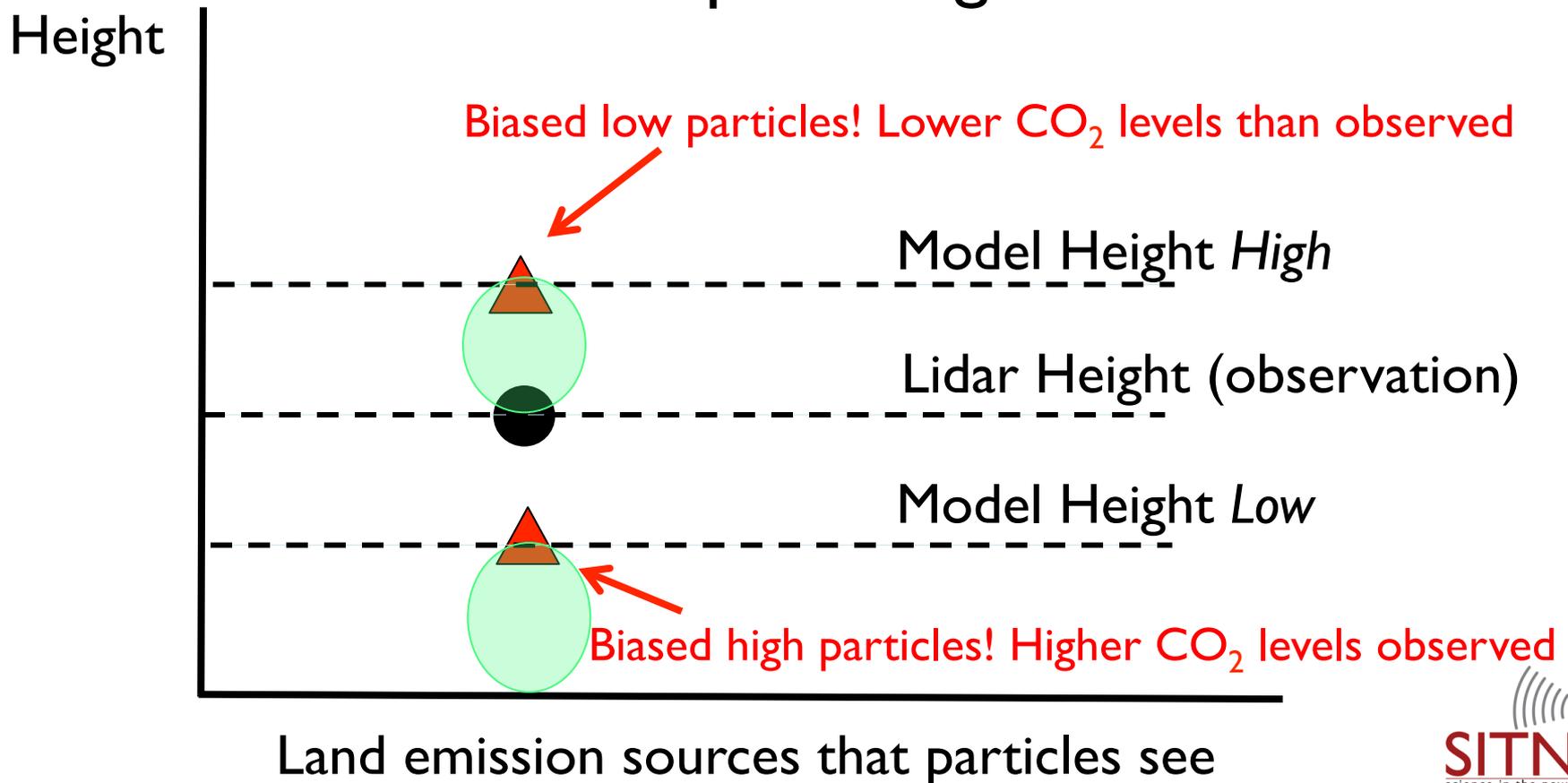
3-D transport of particles



- We can track particles on a horizontal grid (map)
- We can track particles vertically in the atmosphere and see what air pollution source most influence CO₂ levels!

Particles may be biased low or high to land emission sources...

If we do not model the height of the lowest layer of atmosphere right!



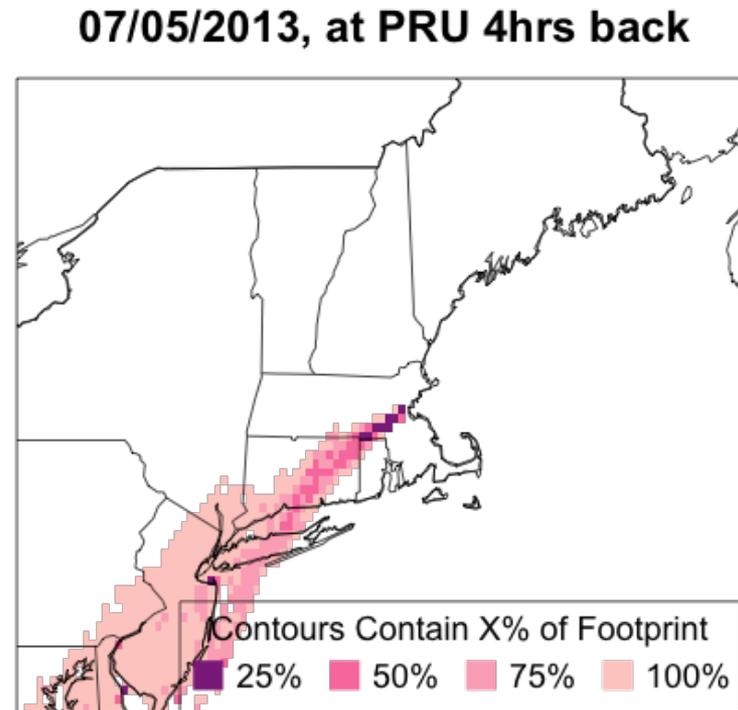
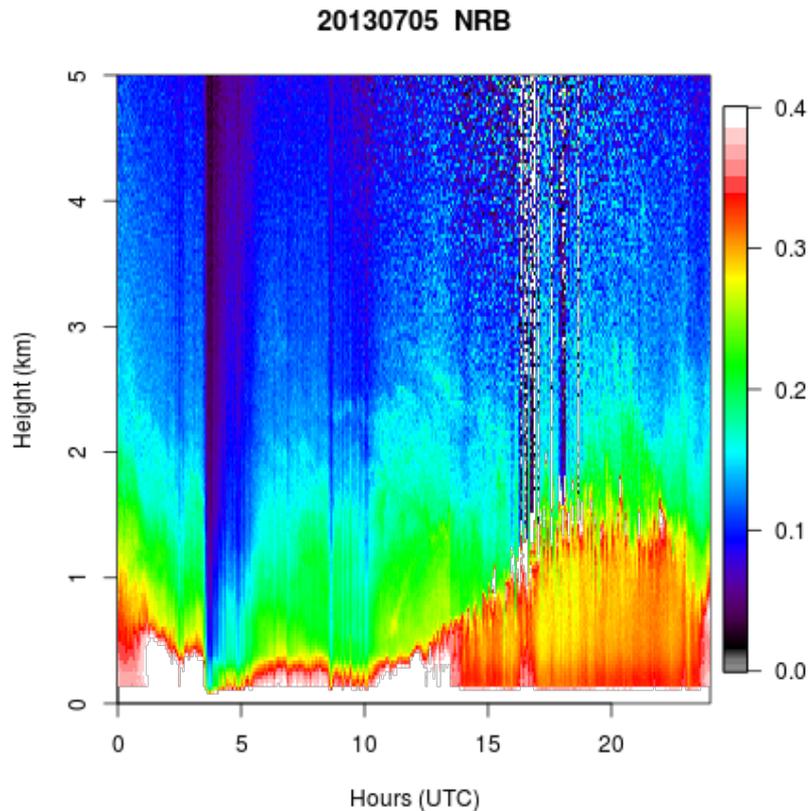
Develop Framework for Identifying “Good Dates” for Transport Modeling

- If Lidar measurements do not agree with transport model
X → Do not use/rely on these dates and model results
- If Lidar measurements do agree well with transport model
✓ → Study transport of air pollution in cities
- Framework used to identify uncertainties in model results!

Let's look at an example of a “good date” to perform this type of modeling.....

What does it look like when we track particles in Boston?

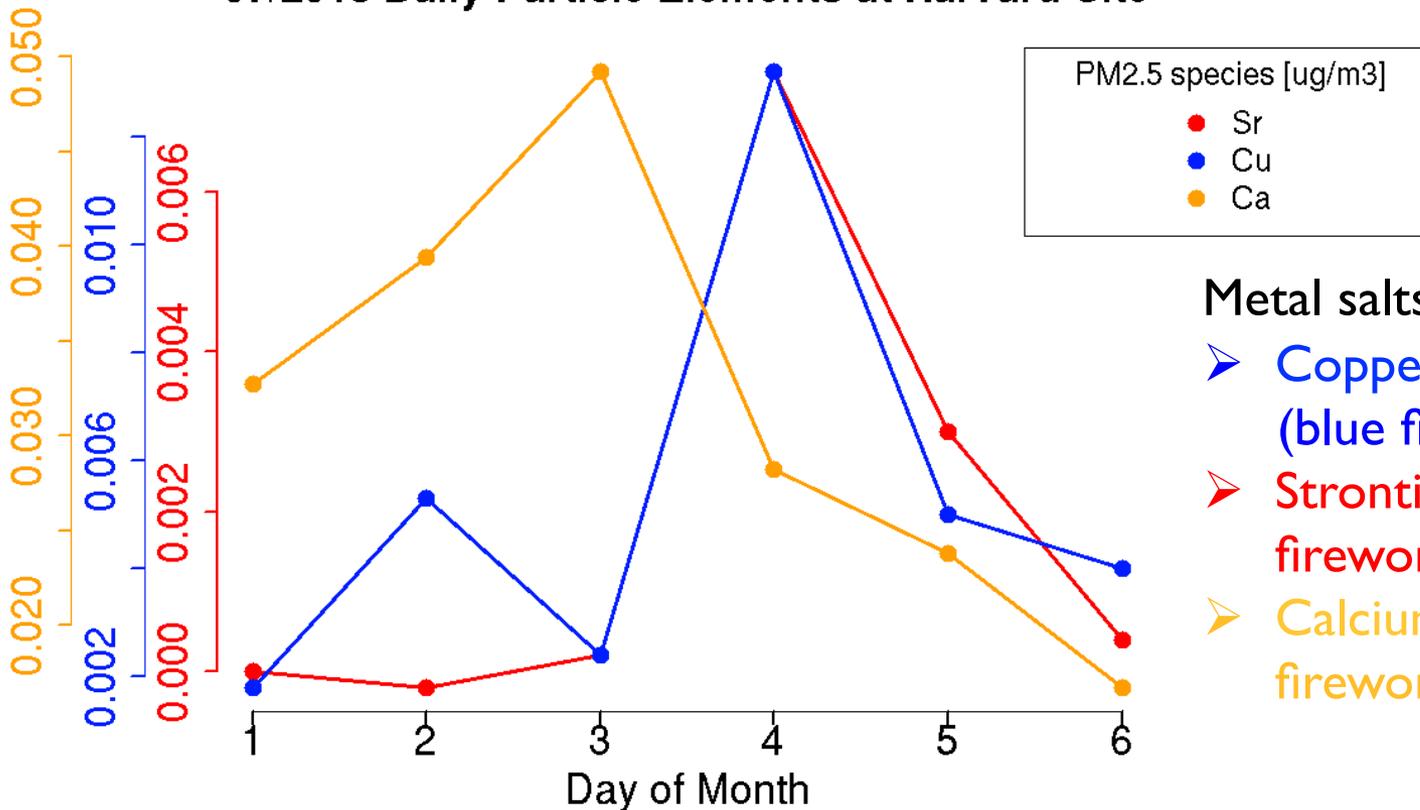
Can you guess a major source of particle pollution?



July 4th Fireworks!



07/2013 Daily Particle Elements at Harvard Site



Metal salts commonly used:

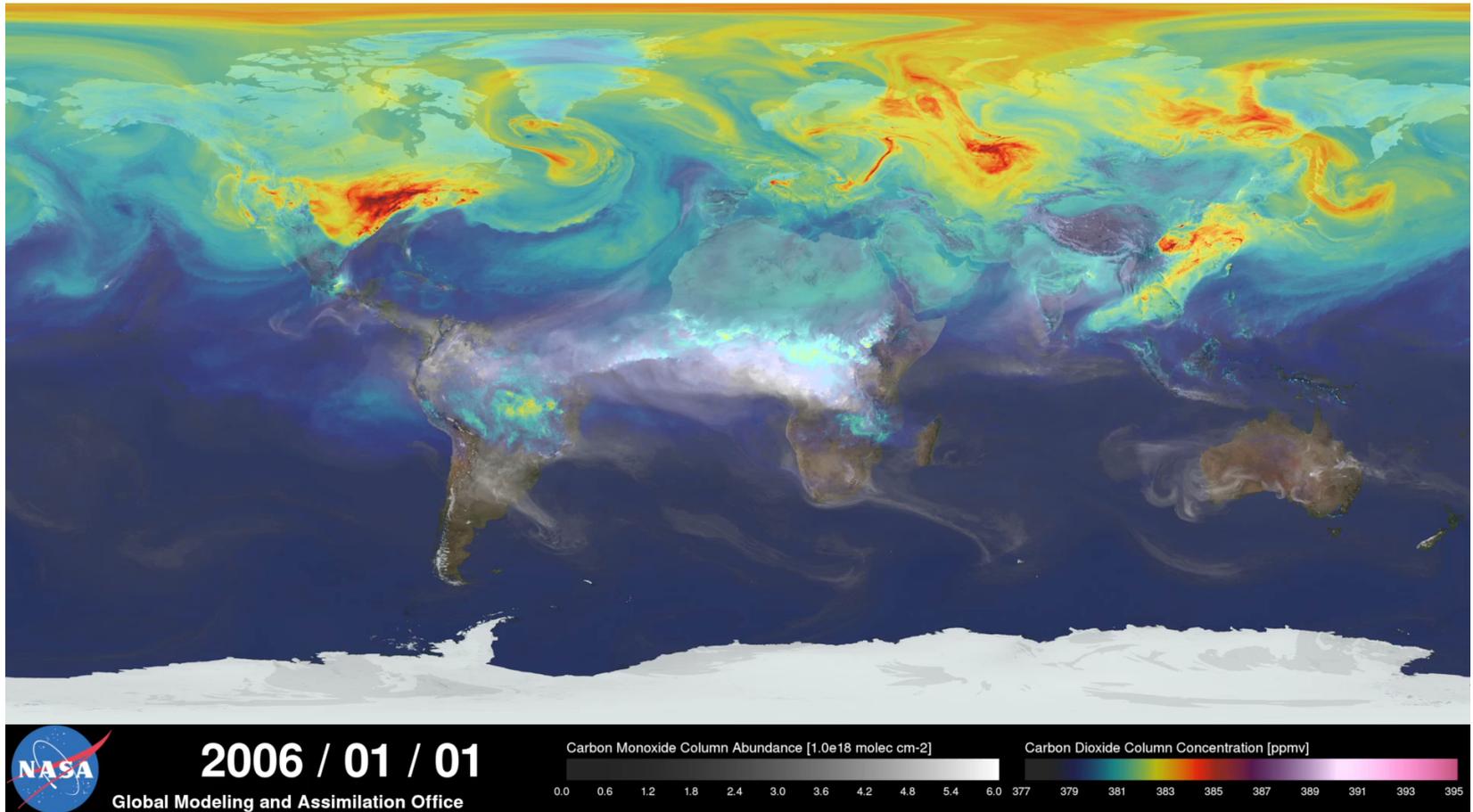
- Copper chloride (blue fireworks)
- Strontium carbonate (red fireworks)
- Calcium chloride (orange fireworks)

Pause for Thought

- How much CO₂ emissions will the carbon cycle continue to absorb in the presence of climate change?

We don't know.....we must keep track of these changes on a global scale!

Global Simulation of CO₂



Conclusion

- Atmospheric scientist use various high-precision instrumentation networks to understand levels of air pollution
- The carbon cycle is complicated so these observations do not tell us the whole picture!
- We need models to incorporate complex processes of the carbon cycle, and combine information from industrial emissions inventory, as well as those from the biosphere
- Air pollution knows no boundaries! We need to understand the transport and places where we as humans can reduce energy-related carbon emissions
- The Paris Climate Agreements calls for us to understand the areas in which cities, states, and countries can improve its energy infrastructure

Thank you!

SITN would like to acknowledge the following organizations for their generous support.

Harvard Medical School

Office of Communications and External Relations
Division of Medical Sciences

The Harvard Graduate School of Arts and Sciences (GSAS)

The Harvard Graduate Student Council (GSC)

The Harvard Biomedical Graduate Students Organization (BGSO)

The Harvard/MIT COOP



<https://sitn.hms.harvard.edu>



SITNBoston@gmail.com



[@SITNHarvard](https://twitter.com/SITNHarvard)

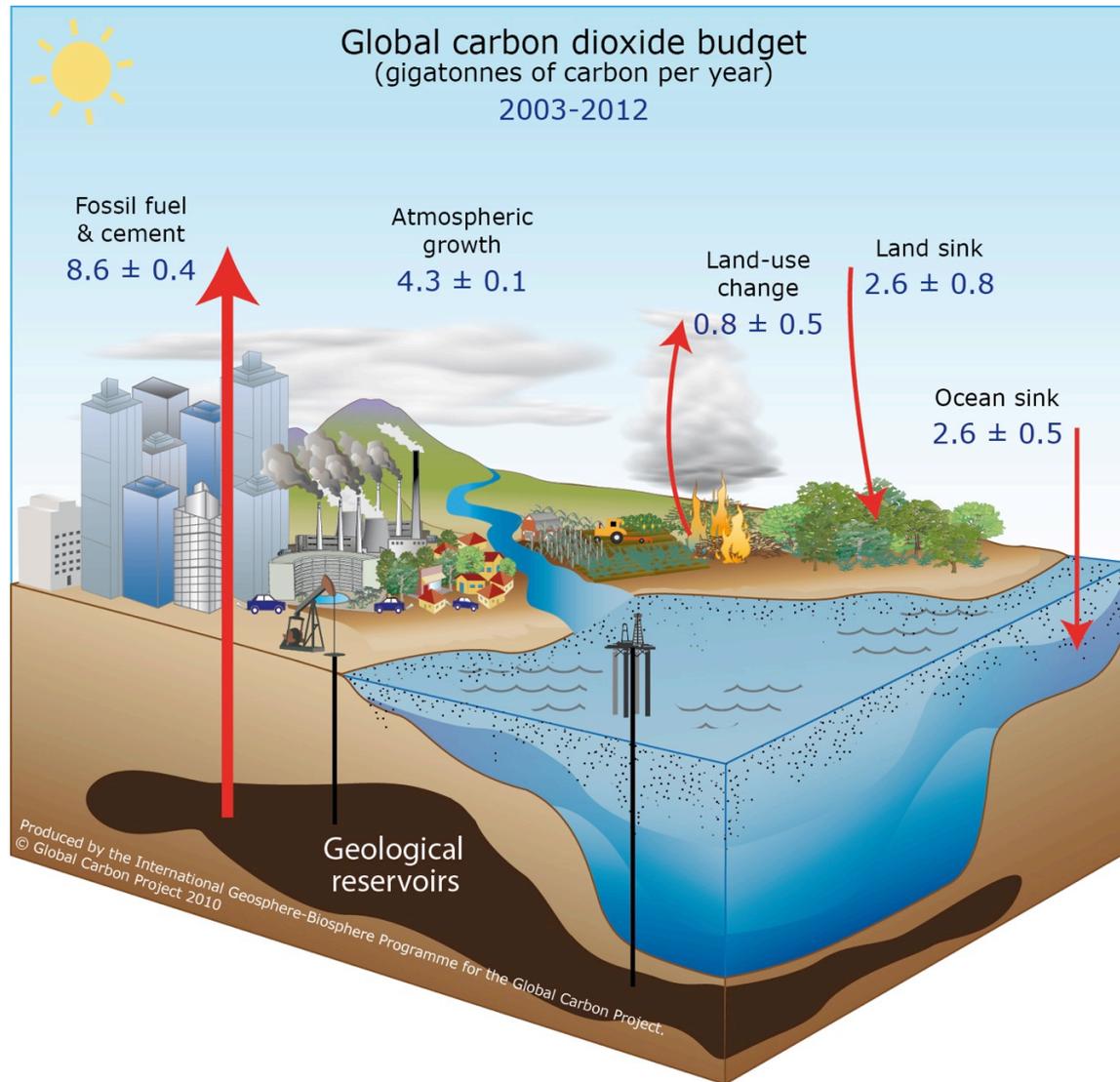


Like

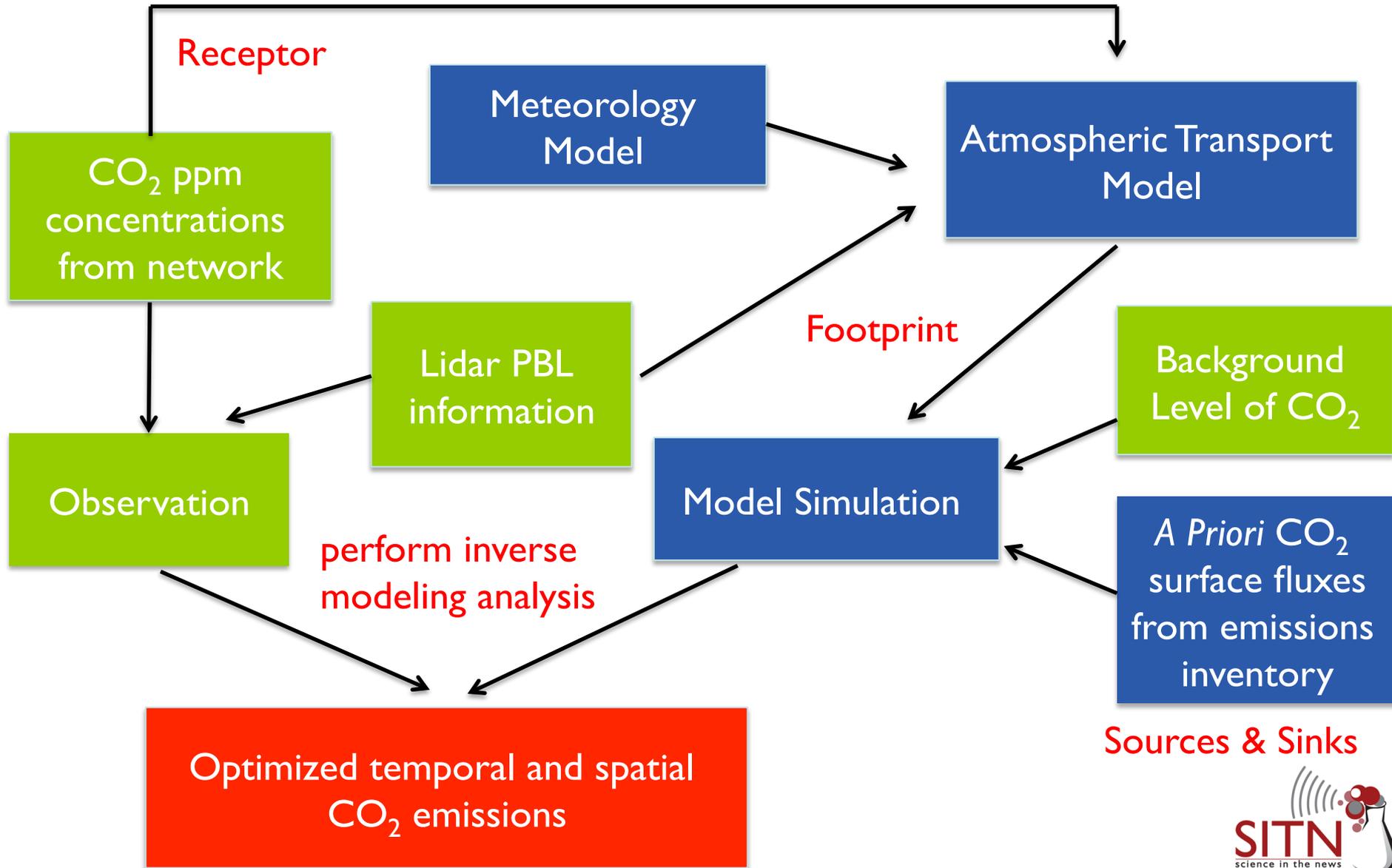
Facebook.com/SITNBoston



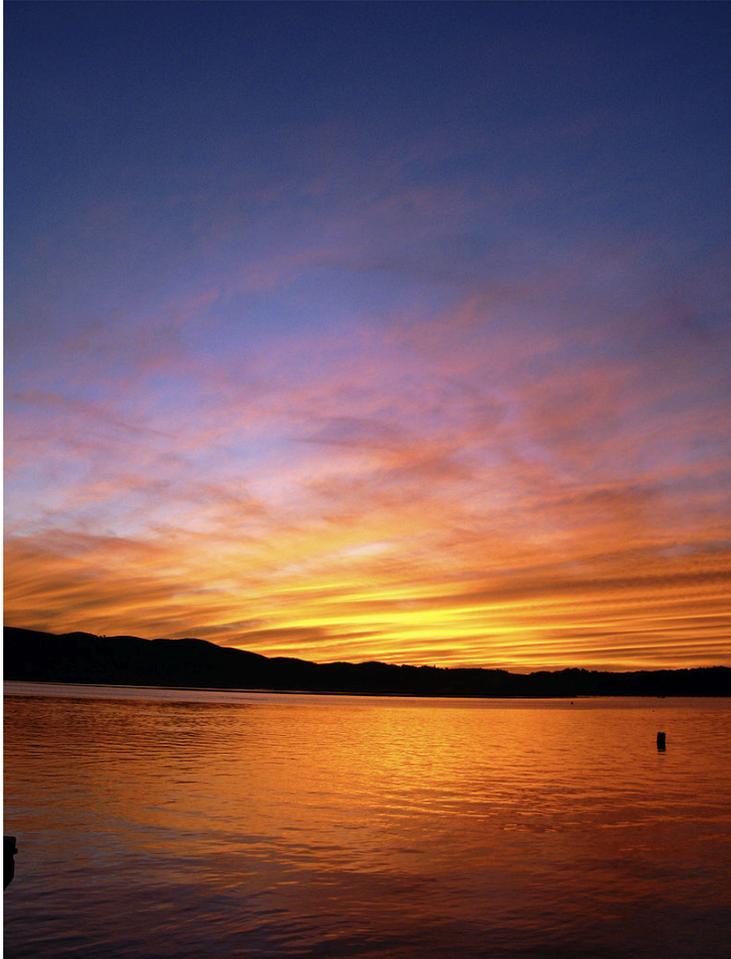
The Carbon Budget



Modeling Framework

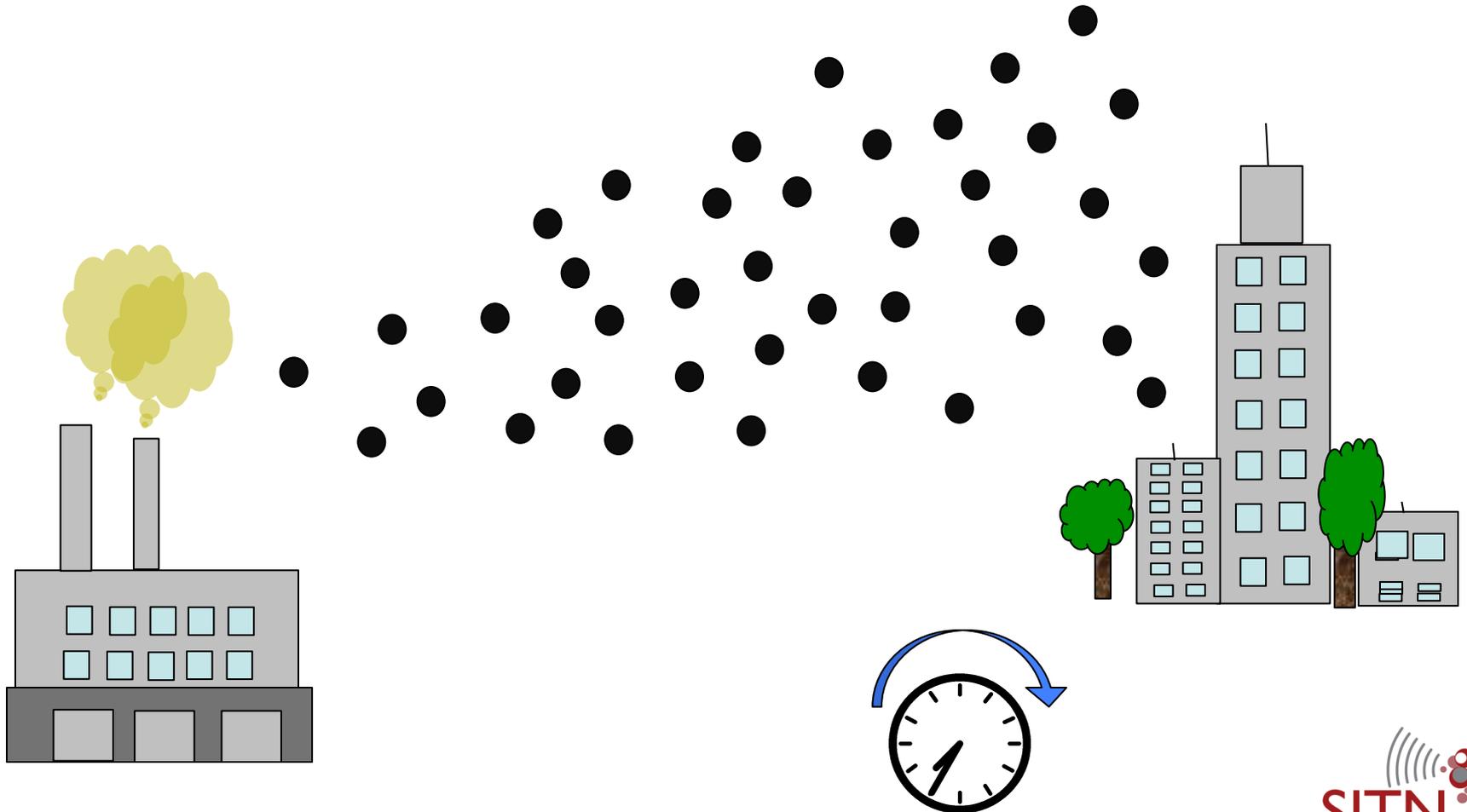


Scattering and backscattering



- Atmospheric gases scatter the red/orange colors at sunset
- Aerosols (suspended particles), haze, and fog, scatter more and that's why we see the grey/white colors in the background and in clouds
- Backscattering deflection of light particles at an angle of 180°

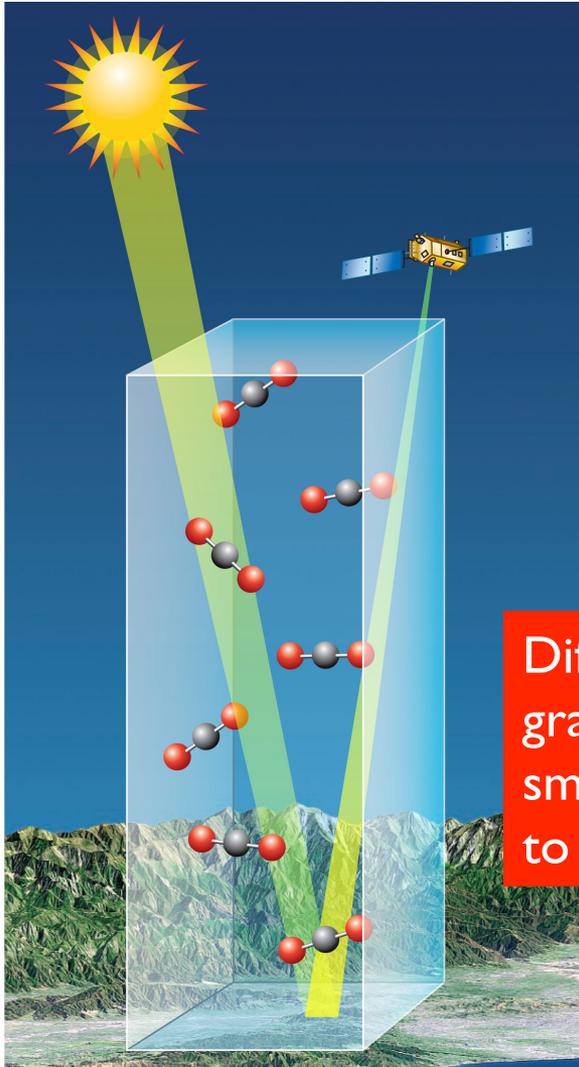
We get these products by using a transport model



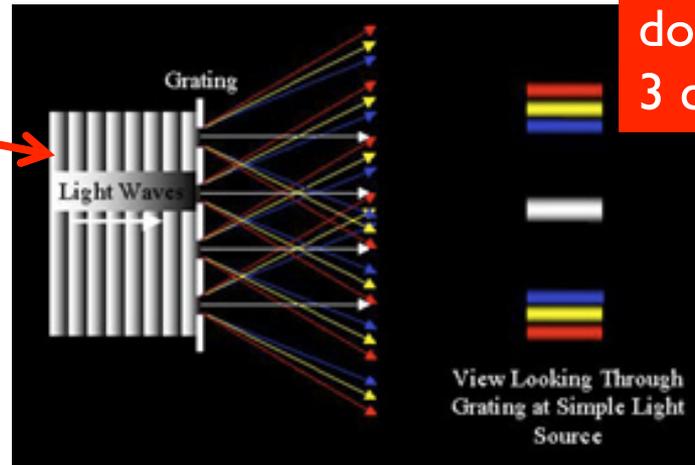
Forwards in time

How does OCO-2 work?

- Measures the intensity of light reflected from the sun on air volume containing air pollution V
- What makes OCO-2 different from instruments used on ground?
 - They can measure up to 17,500 different colors so any shift in light intensity provides higher precision of estimating CO₂
 - CO₂ absorbs light at specific colors



Diffraction grating with small grooves to spread light



Your camera does this using 3 colors!